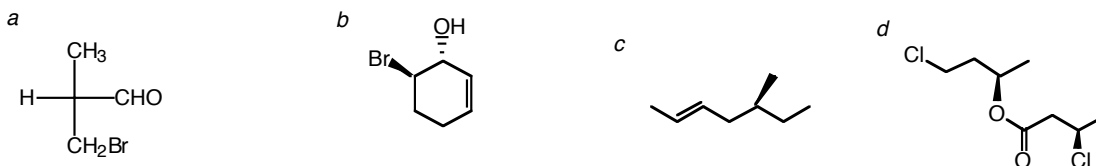
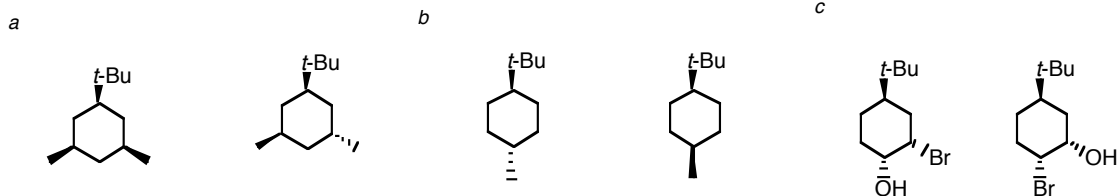


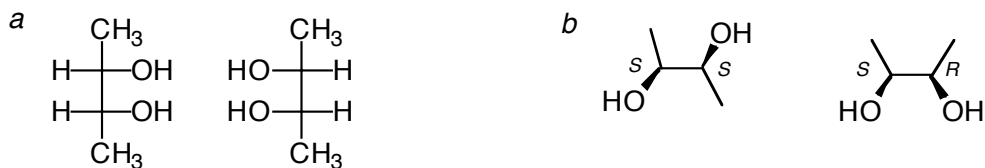
1. Assign all stereochemistry (R or S; E or Z)



2. For each pair, which is more stable?



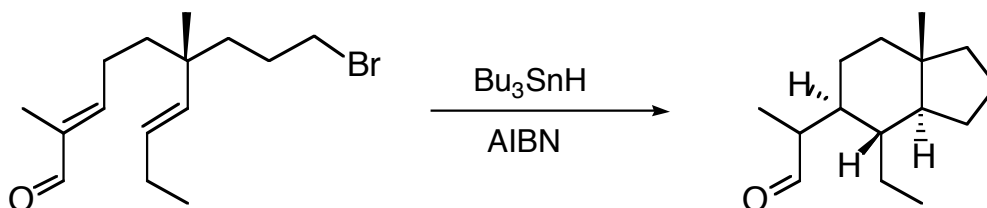
3. Indicate the relationship for each pair below: enantiomers, diastereomers, or meso



4) For each pair, indicate which is more stable. Use a clear picture to explain why



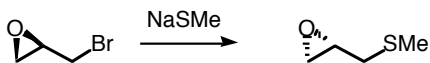
5. Provide a detailed arrow pushing mechanism for the following reaction



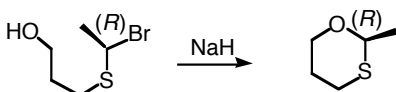
6. Draw Newman projections for the staggered and eclipsed forms of n-butane. Give the relative energies of all of the eclipsed and staggered conformers.

7. Gauche butane is  $\sim 0.8$  kcal/mol higher in energy than anti butane. The A-strain of a methyl group on cyclohexane is  $\sim 1.7$  kcal/mol. Use Newman projections to explain the relationship in detail.

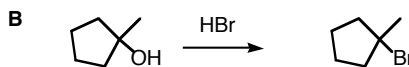
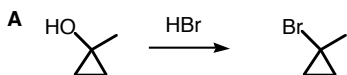
8) Provide a mechanism



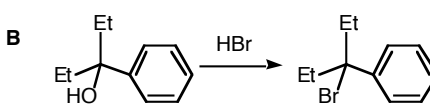
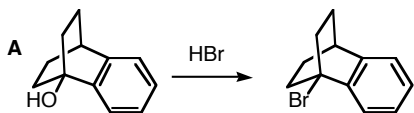
9) The reaction below is an  $S_N2$  process. However, the absolute configurations of the starting material and products are both R. Explain. (Hint: this is a trick question).



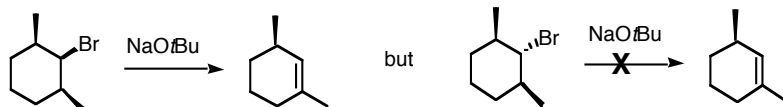
10) Which reaction is faster (**A** or **B**)? Why?



11) Which reaction is faster (**A** or **B**)? Why?



12) Provide a detailed explanation



13) Provide a detailed arrow pushing mechanism.

