

UNDERSTANDING BIOGEOCHEMICAL TRANSFORMATION AND MOBILIZATION OF HG FROM RIVER BANK SOILS: SOUTH RIVER, VIRGINIA



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INTRODUCTION

ELAWAR

Between 1929 and 1950, mercury (Hg) was extensively used by the DuPont plant in the production of rayon acetate fiber in Waynesboro, VA and released into the South River. The contamination of Hg was discovered in the 1970s and remains elevated in water, soil, sediments, and biota.

RESULTS

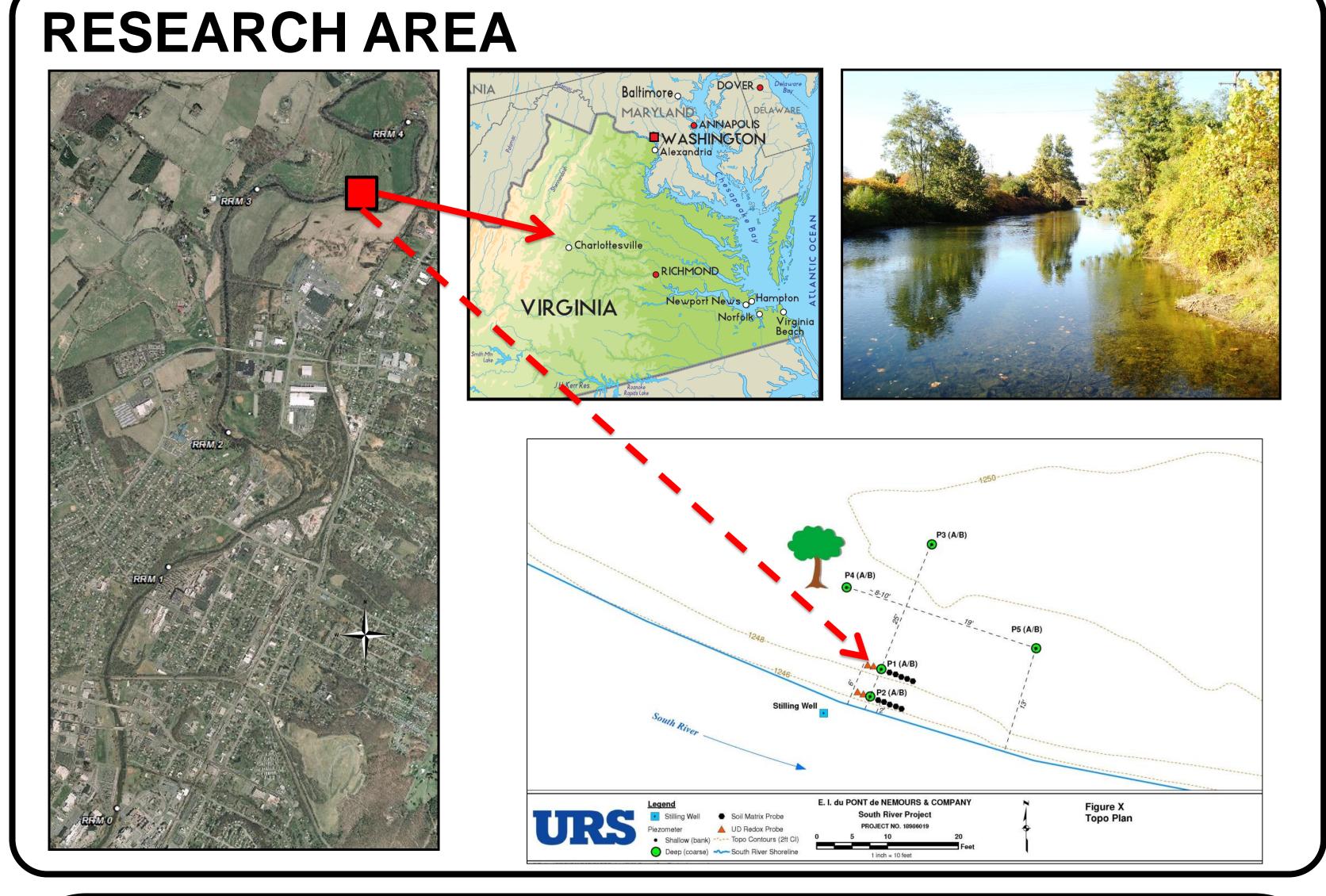
CORE 2: 0.6 m from bank

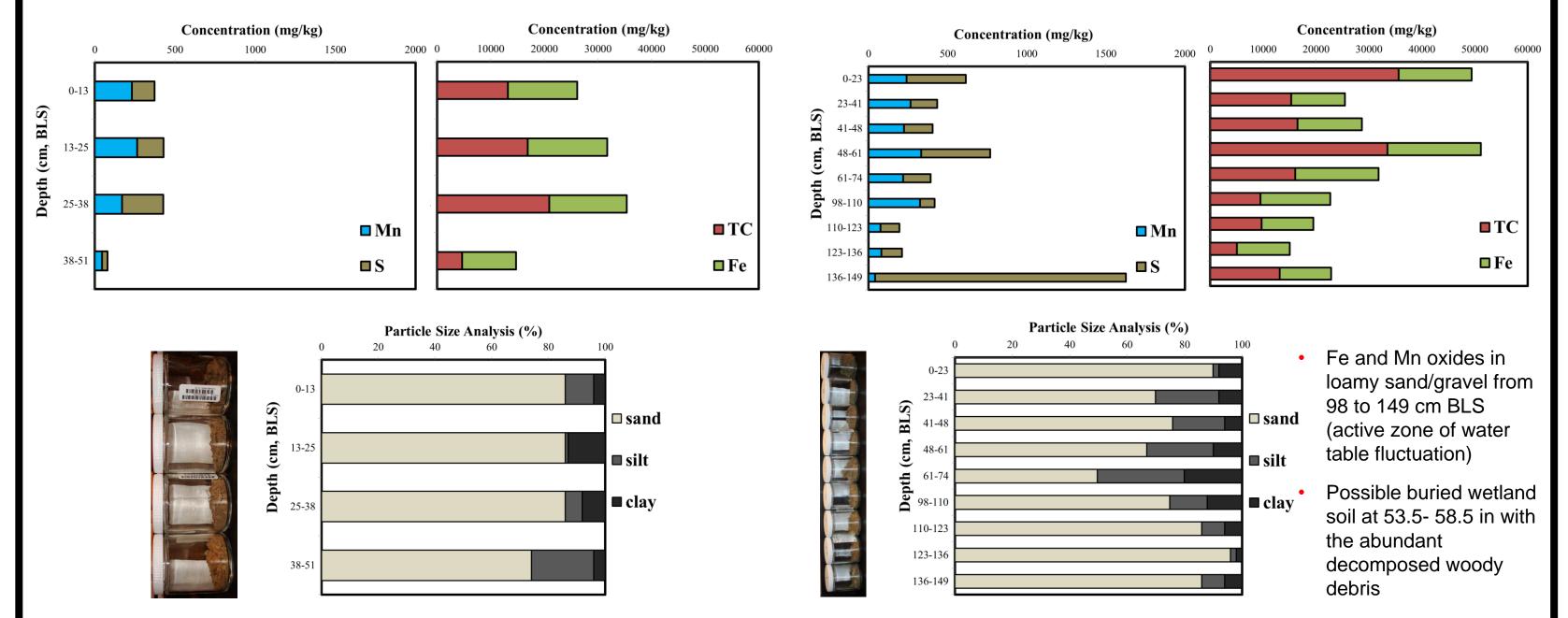
CORE 1: 1.8 *m* from bank



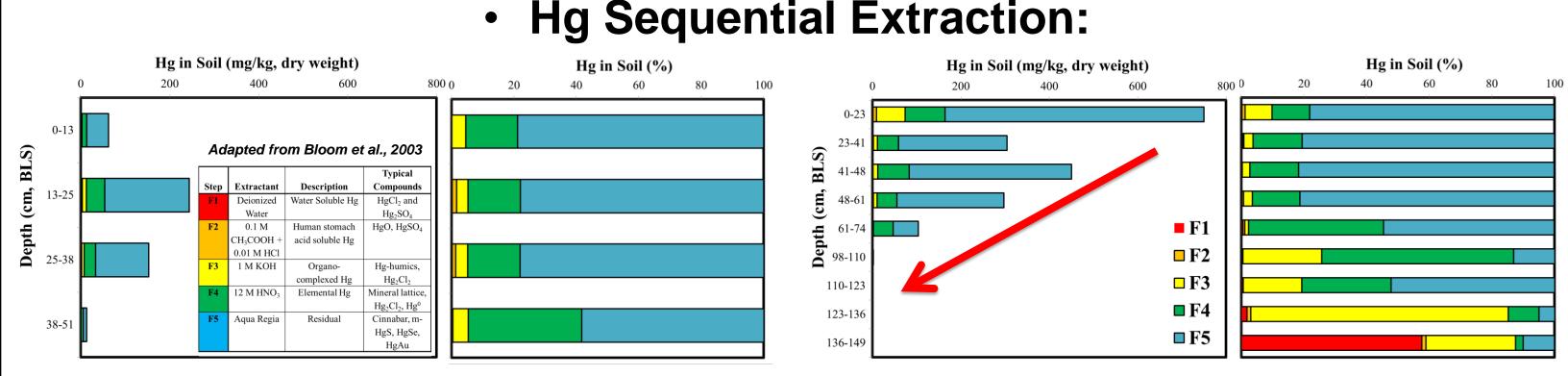
The major objectives of this study are to investigate the processes that govern biogeochemical transformation and mobilization of Hg from the floodplain soils at South River Mile 3.5, understand geochemical gradients within river bank soils and how they change over time using sensor technology, and characterize how Hg cycling and transport is affected. Our over-arching hypothesis is to test if leaching of bank soils is a significant source of dissolved inorganic Hg (IHg).

Our major hypotheses are: (1) soil inundation of bank soils are due to horizontal flow through a highly transmissive gravel zone at the base of the bank, vertical drainage of precipitation, and upgradient groundwater flow; (2) drainage occurs predominantly through gravel zone wetting an organic rich soil; and (3) hydraulics facilitate the downward or upward movement of the capillary fringe affecting soil redox potential, mineral dissolution, and leaching of inorganic Hg into dissolved/colloidal phases that are either directly transported to the river or methylated within the saturated zone of the bank and subsequently released.





Particle size analysis showed that soils were dominated by the sand fraction (50-90%) with the variable percentage of silt (1-30%) and clay (2-20%) fractions, respectively



Concentration of total Hg in core 1 was up to 750 mg/kg (dry weight) dramatically decreasing with depth to 0.1 mg/kg! Hg concentration in a core 2 was up to 244 mg/kg and could be lower due to a closer proximity to the stream flashy vents. Core 1: only one sample from 136-149 cm BGS (wetland soil) had 58 % of water soluble Hg; about 1% of weak acid extractable Hg was in all samples, whereas organo-complexed Hg increased with depth (up to 82% at 123-136 cm BGS); strong complexed Hg was the highest between 61 and 110 cm (up to 61%); residual Hg decreased with depth from 80 to 5%. Core 2: residual Hg is the major component in all samples (about 80%) while Hg generally increased with depth (up to 36%)

METHODS

Installation of in-situ real-time monitoring sensors





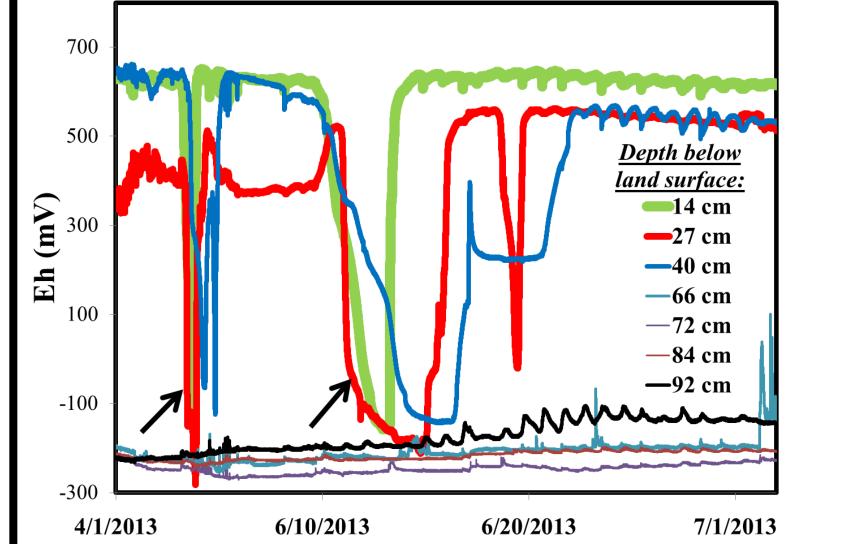
Soil moisture, *temperature (Decagon)*

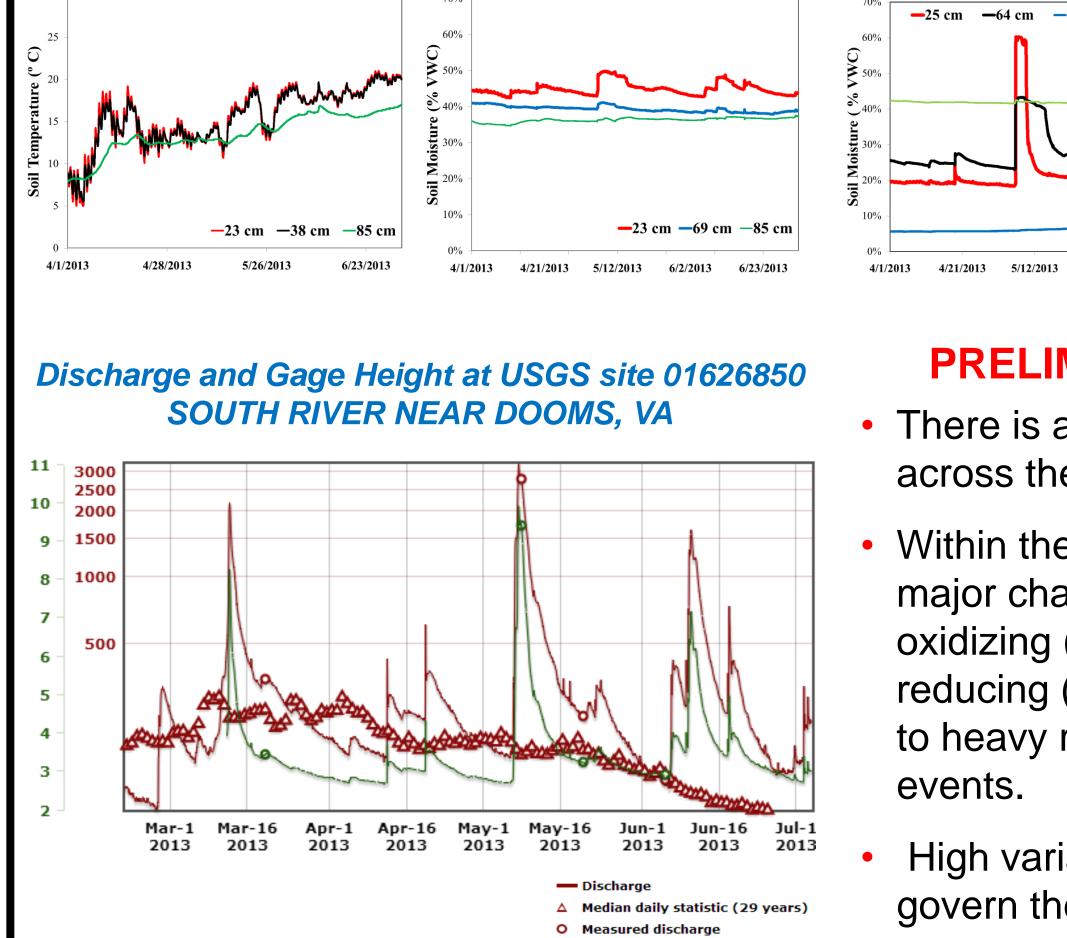


Sampling and analysis of soil cores, stream water, and shallow groundwater

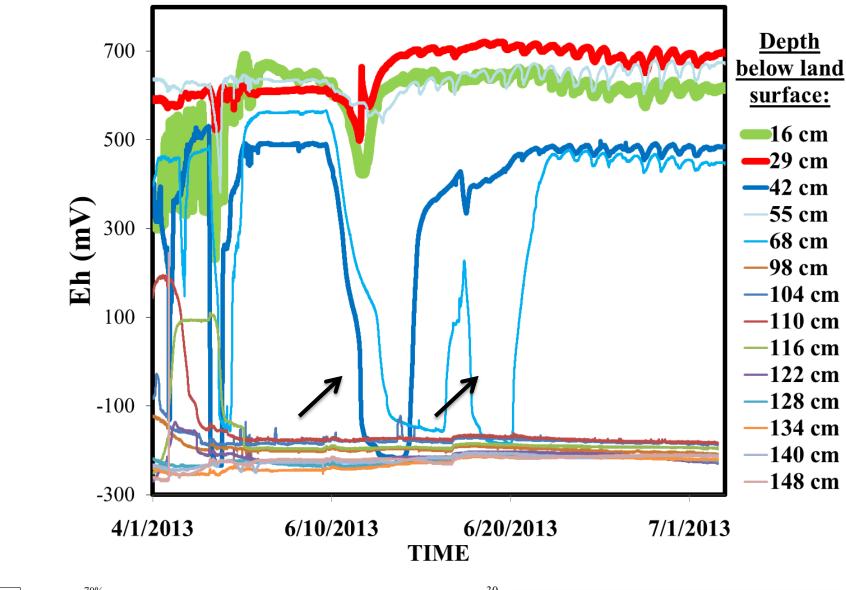
IN-SITU SENSOR DATA:

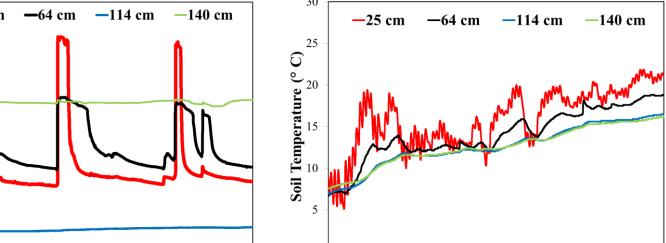
The sensors are used to characterize geochemical gradients and how they change over time, and to enable targeted sampling at Hg loading hot spots

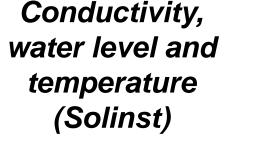




Measured gage height











The chemical analysis included the following: pH; cation exchange capacity plus exchangeable cations (K, Ca, Mg, Na); total carbon and total nitrogen; particle size analysis; ICP analysis of digestates (Fe, Mn, Al, S, Si); DCB extraction for Fe oxides, total Hg, methyl Hg and sequential extraction Hg analyses.



PRELIMINARY CONCLUSIONS:

- There is a significant redox gradient across the floodplain soil profile.
- Within the top 40 70 cm of Hg-rich soil, major changes in redox conditions from oxidizing (Eh ≈+600 mV) to very reducing (Eh ≈-300 mV) corresponded to heavy rainfall and overbank flooding
- High variations in stream stage may govern the surface water - groundwater exchange affecting soil redox potential, stability of Hg-bearing minerals and mobilization of Hg.