Instructor: Dr. J. A. Wingrave (315 QDL, 831-1676) (wingrave@udel.edu)

• Website for CH437: (www.udel.edu/chem/wingrave/chem437/chem437.html)

Required Course Supplies (Available at University Bookstore):

• **Textbook**: Skoog, Douglas A., Holler, F. James, Nieman, Timothy A.

Principles of Instrumental Analysis, 5th Ed.

• Solutions Manual: Skoog, Douglas A., Holler, F. James, Nieman, Timothy A.

Solutions Manual to Accompany, Principles of Instrumental

Analysis, 5th Ed.

## Other Resources

- Tutors For more information, see Mrs. MacMillan in BRL 104
- Office Hours: 2:30-4:00 pm MW & 7:30-9:00 am F in 315 QDH
- Bookstore Chemistry flash cards, Schaum Outlines, Chemistry Problem Solvers, etc.

## **GRADING POLICIES**

## The minimum requirements for obtaining a passing grade in CH-437 are:

- Successful completion of the final examination
- Obtaining a total of atleast 100 points on the grading scheme outlined below

There will be 300 points that can be earned in CH-437 from the sources below;

- Examinations ( 200 points, 50 %) Three (3), 100 point exams will be given but only the highest two (2) exam scores will count toward your grade total. Your lowest exam score will be dropped. For missed exams a score of zero (0) will be recorded: NO make up exams will be given for any reason. NOTE: Exam #3, will be given at the same time as the Final Exam.
- Final Examination (100 points, 50 %) The final exam will be cumulative & exam #3, will be given at the same time as the Final Exam.
- Grading Schedule for CH-437

TOTAL POINTS	<u>GRADE</u>	TOTAL POINTS	<u>GRADE</u>	TOTAL PO	<u> STNIC</u>	<u>GRADE</u>
300-265	Α	215-200	В-	150-135	D +	
265-250	A -	200-185	C +	135-115	D	
250-235	B +	185-165	С	115-100	D -	
235-215	В	165-150	C -	99-0	F	

## CHEMISTRY 437 Wingrave -- Lecture, Lab & Exam Schedule -- Spring 2000

DATE	T LEOTUDE TODICO	OTUDY
DATE	LECTURE TOPICS	STUDY PROBLEMS
Feb. 7,9	CH 1. Syllabus, Measurement, Analysis & Calibration	1: 9,10; TABLE 1-3 Appendix 5
Feb. 14, 16	CH 2-5. Electronics & Noise	<b>2:</b> 5,11,12,18; <b>3:</b> 1; <b>4:</b> 1-4,6; <b>5:</b> 1-12;
Feb. 21, 23	APPPENDIX 1. Error Analysis	Appendix. 1:1-21 (s or σ, by calculator)
Feb. 28, March 1	CH 6-8. Intro to Spectroscopy & Spectrometers	<b>6:</b> 1-5,9,11,12,14-17; <b>7:</b> 1-3,6,8,15-17, 21-23; <b>8:</b> 2,3,6,8,9
March 6	EXAM #1	CH 1-8, Appendix #1
March 8	CH 9-12,20. Atomic & Mass Spectroscopy	<u>9:</u> 1,,3-6,8-12,(21,22);   <u>10:</u> 1-10;
March 13, 15	CH 13-16. UV, Vis and IR Spectroscopy	<b>13:</b> 1-10,13-24,29,30; <b>14:</b> 4-7,9; <b>15:</b> 1-5,7; <b>16:</b> 1,3,4,7,8,12
March 20, 22	CH 17-19. IR, Raman & NMR & Spectroscopy	<b>17:</b> 2-8; <b>18:</b> 1,2,5,7 <b>19:</b> 1-12,21-42
March 25- April 2	SPRING BREAK!!	SPRING BREAK!!
April 3	CH 21. Surface Spectroscopy & Microscopy	<b>21</b> : 1-9
April 5	Exam #2	CH 9-21
April 10,12	CH 22. Electroanalytical Chemistry	<b>22</b> : 1-17
April 17, 19	CH 23. Potentiometry	<b>23:</b> 1-15,17,18,20,21
April 24, 26	CH 24, 25. Coulometry & Voltametry	<b>24:</b> 1,2,4-7 <b>25:</b> 1-9,11
April 24,26	CH 26. Chromatography Fundamentals	<b>26:</b> 1-25
May 1,3	CH 27, 28. HPLC & Gas Chromatography	<b>27</b> : 1-19,22-25
May 8,10	CH 28. Ion Exchange Chromatography	<b>28:</b> 1-22
May 15	CH 29, 30. Supercritical Fluid & Capillary Chromatography	<b>29:</b> 1-7 <b>30:</b> 1-6
May 17	CH 31, 32. Radiochemical & Thermal Analysis	31: 1-11 32: 1-10,12-21
May 18	Reading Day	
To Be Announced	EXAM #3 & FINAL EXAM CH 1-32 & Appendix #1	

WINGRAVE Chemistry 437	7 Personal Information Form	Spring, 2000			
Please complete this form and return it	to Professor Wingrave at the end	of the first class.			
Name:	Section:				
UD e-mail Address:	Phone:				
At UD: Major:	Minor:				
<ul> <li>Previous Science and Math Courses: at UD write "UD" or; 3) more than 5 year</li> </ul>	If you took the course in; 1) High ars ago write "5", in the blank.	School write "HS"; 2)			
Algebra I	Algebra II	Geometry			
Trigonometry	Geometry	Pre-Calculus			
Calculus	Differential Equations	Statistics			
Quantitative Analysis	Organic Chemistry	Analytical Chem.			
Inorganic Chemistry	Physical Chemistry	Biology			
Chemical Engineering	Enviromental Chemistry	Biochemistry			
Other Chemistry (list)		·			
Other Engineering (list)					
• The role of the instructor in CH-437 [F	Rate from 1 (most) to 5 (least) impo	ortant / desirable]:			
how to work chemistry an instrumentation and operate the chemistry you will nee	w the principles upon which equati d science problems,	classes?			
<ul> <li>Instrumental methods of most interes</li> </ul>	st;				
Electrical Circuits	Error & Noise Analysis	Mass Spec			
Atomic Adsorption Spec.	Atomic Fluorescence Spe	c UV-Vis Spec.			
Radiochemical Analysis	Nuclear Magnetic Resona	ince IR Spec.			
Electrochemical Analysis	Surface Analysis	Raman Spec.			
Thermal Analysis	Gas Chromatography	HPLC			
Capillary Electrophoresis	Supercritical Fluid I	Extraction			
Other Methods (list)		•			