CHEM-643 Biochemistry Mid-term Examination 8:00 – 10:00, Monday, 25 October 2004	Name
Dr. H. White - Instructor	
Read every question so that you underst	uding this page. Write your name on each new page. nd what is being asked. If you feel any question is answer or interpretation. Please call my attention to
	hay refer to your assignments and your lecture notes, the metabolic pathway sheets available from the course
clearly. It is intended to be challenging	g, problem-solving skills, and ability to communicate even to the best students in the class. Some of the not seen before and is not in your text; however, the ic principles discussed in the course.
Do not expose your answers to the scrut before you go on to the next.	ny of your neighbors. Please fold under each page
Breakdown of the examination by sections	
I. Short Answer II. Natural Product S III. Problems IV. <u>Short Essays</u> Total	ynthesis 15 Points 15 Points 50 Points 20 Points 100 Points
Exam Statistics	Class Range

Class Mean

_____Your Rank in class

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	(1	
Part I - Short Answer Questions (1 point each)		

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Part I - Short Answer Questions (point each)
1.	Name of the bicyclic ring system found in folic acid and several other cofactors.
2.	Antidote for ethylene glycol poisoning.
3.	Class of PLP-dependent reactions that have a "ping-pong" mechanism.
4.	NADH in the presence of oxygen in the mitochondria is used to make ATP via this process.
5.	Name of the vitamin precursor to FAD.
6.	This cofactor helps catalyze carbon-carbon bond cleavage next to carbonyl groups.
7.	Virtually exclusive metabolic fuel for your brain.
8.	Biosynthetic precursor that provides all of the carbon in 2,4,6-trimethyl nonanoate in the dipper duck.
9.	Example of a fatty acid that is a diterpene.
10.	Enzyme in <i>E. coli</i> that is feedback inhibited by histidine, AMP, GMP, glycine, alanine, and glucosamine-6-P.
11.	Fatty acid precursor of prostaglandins.
12.	Group of organisms that do not make steroids.
13.	Golden Rice was engineered to make this compound.
14.	Amino acid source of methyl groups in methylation reactions.
15.	Small FeS-containing protein involved in a variety of one electron transfer reactions.

Part II Natural Product Biosynthesis:

1. (5 Points) Homocitrate (below) coordinates to the molybdenum in the FeMo cofactor in nitrogenase. Propose a simple, reasonable enzymatic reaction that would form homocitrate in one step from well-known common metabolic intermediates.

2. (5 Points) The major component of the sex attractant pheromone of the boll weevil, *Anthonomus grandis*, is grandisol, shown below.

- A. (1 Point) To what class of compounds does grandisol belong?
- B. (4 Points) In the space to the left of grandisol, draw the structure of a well-recognized, likely precursor in a way that shows its carbon-for-carbon relationship to grandisol.
- 3. (5 points) A recent report in *Nature* [429, 363 (2004)] describes the isolation and characterization of a bright red compound in hippopotamus sweat that serves as a sun screen for this virtually hairless tropical animal. The structure of hipposudoric acid is shown below. What compound, displayed in another problem in this examination, would be a likely precursor to this hipposudoric acid and account for all of the carbon atoms in it? Show why you selected the compound you did.

Part III Problems

1. (10 points) Certain molds produce 6-methylsalicylate from the eight-carbon, enzyme-bound intermediate shown below. Show either the reactions required to form the intermediate from acetyl CoA <u>or</u> how mechanistically the intermediate would cyclize and go on to form the product.

2. The final steps in tyrosine biosynthesis in plants and some microorganisms [*Z. Naturforsch* 41c, 69-78 (1986); *J. Bacteriol.* 144, 247-257 (1980)] is shown below. The questions that follow refer to this sequence of reactions.

a) (4 Points) Reaction 1 is the only example known in biochemistry of the Claisen rearrangement, a well-known reaction in organic chemistry. Reaction 3 is an oxidative decarboxylation. Choose one of these two reactions and show by pushing electrons how the reaction proceeds.

- b) (2 Points) What coenzyme is required for Reaction 2? (Give its name rather than an abbreviation)
- c) (4 Points) The coenzyme for Reaction 2 is normally associated with a particular amino acid in the enzyme's active site. How could one chemically link that amino acid residue irreversibly with the coenzyme? (Hint: Problem Set 2) Explain.
- d) (3 Points) What amino acid cosubstrate would you expect to be the amino group donor in Reaction 2? Explain your reasoning?
- e) (5 Points) NADPH is generated in the final step. Does this make sense? Explain your answer.

The catabolic pathway for Phenylalanine and Tyrosine is shown below and is the basis for the questions that follow.

A. (6 Points) What are the names of compounds A and B? What would be the next compounds, C & D, in the metabolism of each?

B. (4 Points) One of the first Inborn Errors of Metabolism, Alcaptonuria, was described by Archibald Garrod over 100 years ago [*Lancet* **2**, 1616 (1902)]. It involved a defect in enzyme 4 above. If [2 ¹³C] phenylalanine were fed to a person with alcaptonuria, where would the labeled carbon be located in homogentisate? Circle the atom in the above figure.

1. (12 points) Knoop in 1905 studied the oxidation of a variety of ω-phenyl acids in rabbits. He fed 2g of each and observed a limited variety in the metabolic products excreted in the urine. His results were as follows.

	Compound Fed	Compound Excreted
1	C ₆ H ₅ -COOH	C ₆ H ₅ -COOH
2	C ₆ H ₅ -CH ₂ -COOH	C ₆ H ₅ -CH ₂ -COOH
3	C ₆ H ₅ -CH(OH)-COOH	C ₆ H ₅ -CH(OH)-COOH
4	C ₆ H ₅ -CH ₂ -CH ₂ -COOH	C ₆ H ₅ -COOH
5	C ₆ H ₅ -CH(OH)-CH ₂ -COOH	C ₆ H ₅ -COOH
6	C ₆ H ₅ -CO-CH ₂ -COOH	C ₆ H ₅ -COOH
7	C ₆ H ₅ -CH=CH-CH ₂ -COOH	C ₆ H ₅ -CH ₂ -COOH
8	C ₆ H ₅ -CH ₂ -CH(NH ₂)-COOH	None observed
9	C ₆ H ₅ -CH ₂ -CO-COOH	None observed
1	C ₆ H ₅ -CH ₂ -CH ₂ -COOH	C ₆ H ₅ -CH ₂ -COOH
0		
1	C ₆ H ₅ -CH ₂ -CH ₂ -CH ₂ -COOH	C ₆ H ₅ -COOH
1		

A. (4 points) Ignoring compounds 8 & 9, explain these results in a coherent way?

B. (4 points) Actually the metabolic products observed by Knoop were glycine amide derivatives of the acids shown above - hippuric acid (C₆H₅-CO-NH-CH₂-COOH) and phenylaceturic acid (C₆H₅-CH₂-CO-NH-CH₂-COOH). Predict the enzyme reaction in which they are formed.

C. (4 points) What is the common name for compound 8? How are compounds 8 and 9 interconverted metabolically? Why didn't Knoop observe metabolic products from them?

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Part V Essay Questions

Writing reflects how you think. Among the "right answers" I will read for the following questions, some will be better than others because they show greater depth of understanding, avoid extraneous or inaccurate information, provide a more logical structure, use appropriate examples, and choose words with precision. Better quality answers will receive higher marks. Therefore organize your thoughts before you write. Strive to write not that you may be understood, but rather that you cannot possibly be misunderstood. Stream of consciousness answers are rarely well organized or clearly presented.

1. (10 Points) Virtually every child born in the United States is tested at birth for a variety of inborn errors of metabolism (genetic diseases). Among them are phenylketonuria, medium-chain acyl CoA dehydrogenase deficiency, biotinidase deficiency, branched chain ketoaciduria, galactosemia, and homocystinuria.

In an informative paragraph or two illustrated with chemical structures, describe one of these diseases. Why is early detection important?

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2. (10 Points) Fix your attention on a carbon atom in a glucose molecule that you consumed at your last meal. In a narrative (no structures), generate a reasonable story of the fate of that carbon atom that involves muscle, liver, brain, and adipose tissue, and several different metabolic pathways before exiting as carbon dioxide.