Ultraviolet Practice #3

Heating ester $\bf A$ in methanol containing sodium methoxide and a trace of water converts it to alcohol $\bf B$ and sodium benzoate $\bf C$. A 1.0 gram sample of $\bf A$ in 50 mL of methanol and excess sodium methoxide shows an absorbance of 0.773 at 293 nm. After adding a little water and heating, the absorbance is 0.600. The molar absorbtivity of $\bf C$ at 293 nm is 3.4. How many milligrams of water have been consumed?