

Understanding Metal Sulfide Growth using ESI FT-ICR MS



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Significance

Making Clusters

Gas Phase Results

Moving Forward

Significance

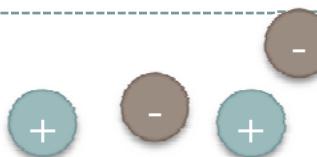


- **Environmental**
 - George Luther: College of Marine Studies
 - Metals in Aquatic Environments
 - ✖ Bioavailability
 - Hydrothermal Vents
- **Materials**
 - Solar Power
 - ✖ CdS

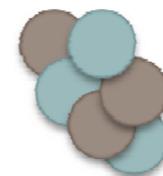
Making Clusters

7T Bruker Apex Qe
FT-ICR MS

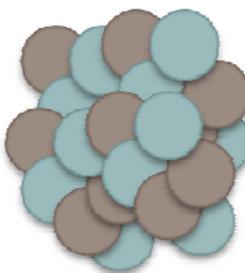
Dissolved Ions



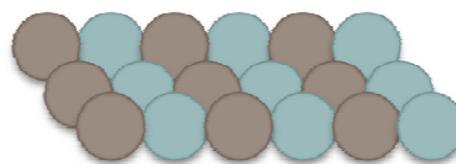
Clusters



Nanoparticles



Solid state material



Size (nm)

1

10

8

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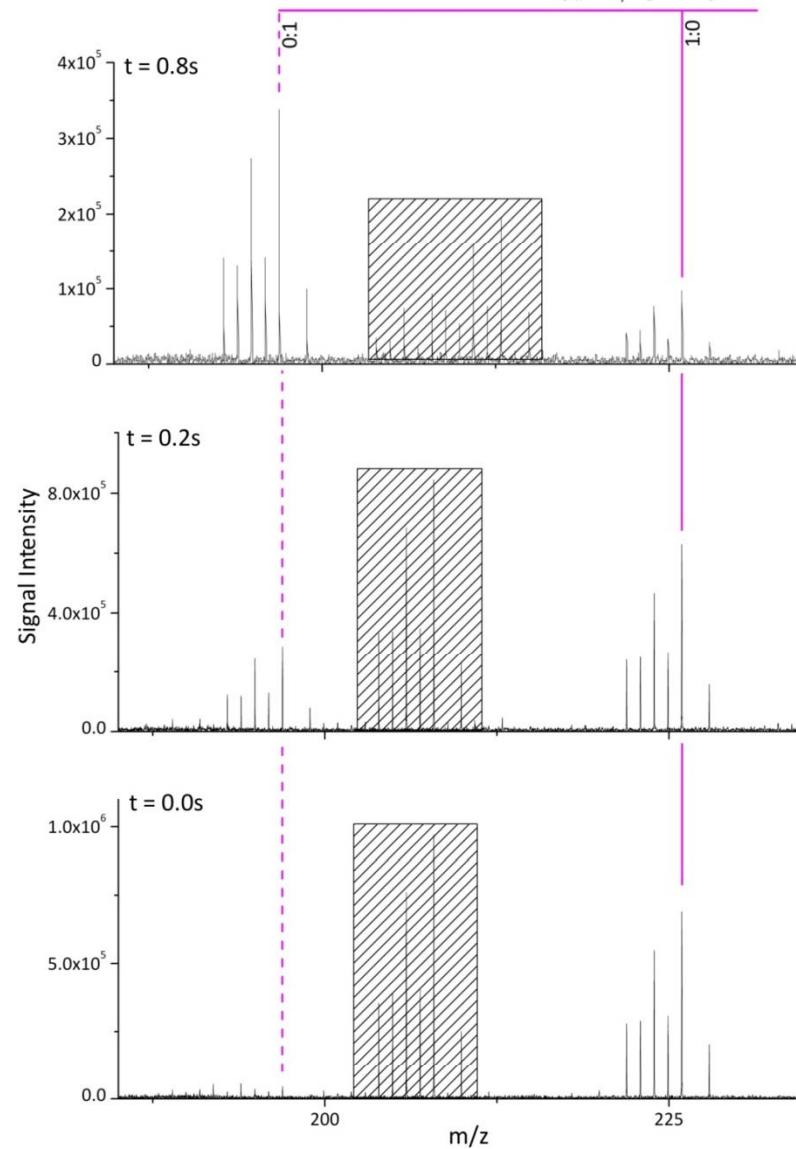
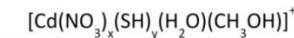
Raw Data



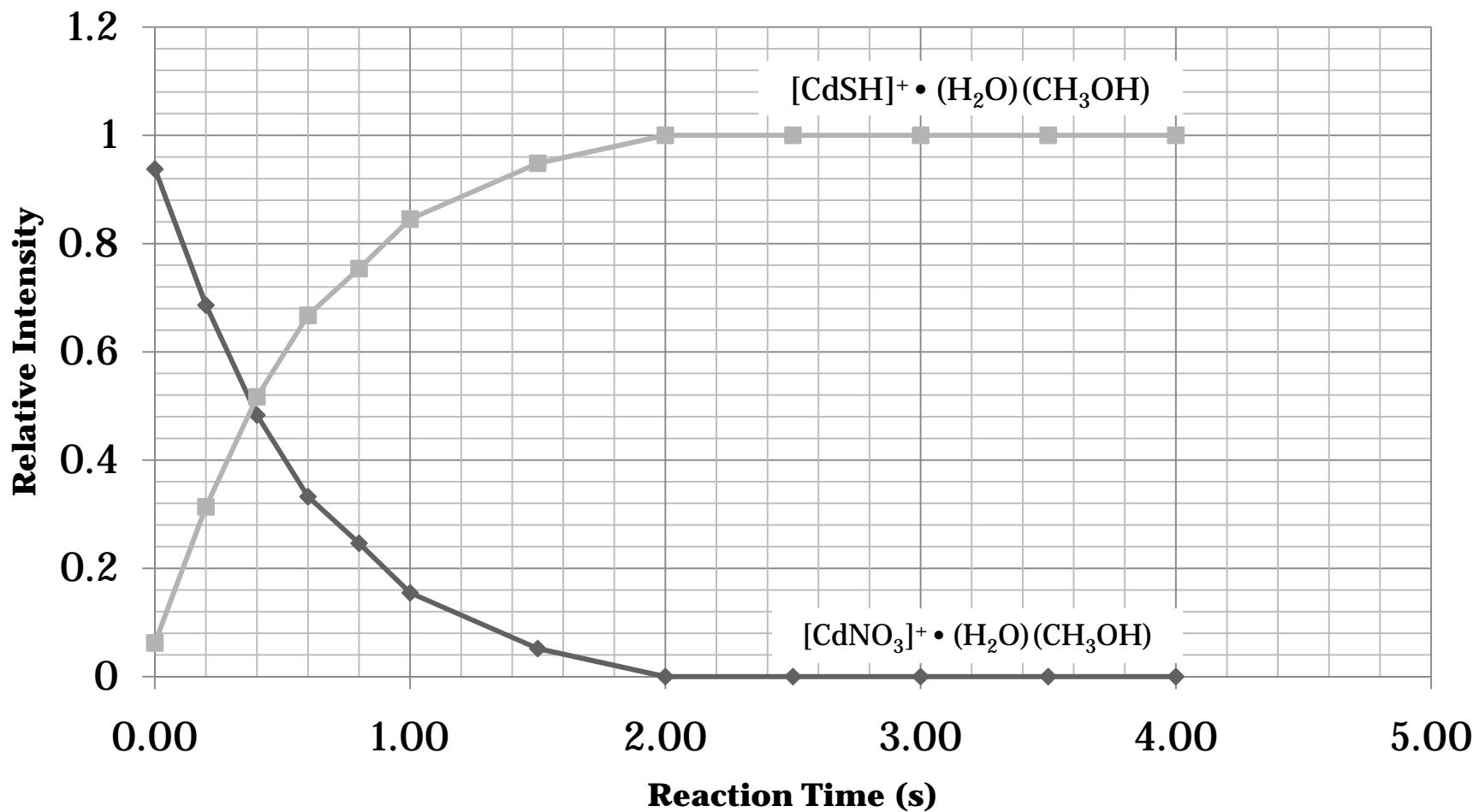
0.5 mM Cd(NO₃)₂ • 4(H₂O)

1:1 water/MeOH

4x10⁻⁹ torr H₂S(g)



Reaction Kinetics



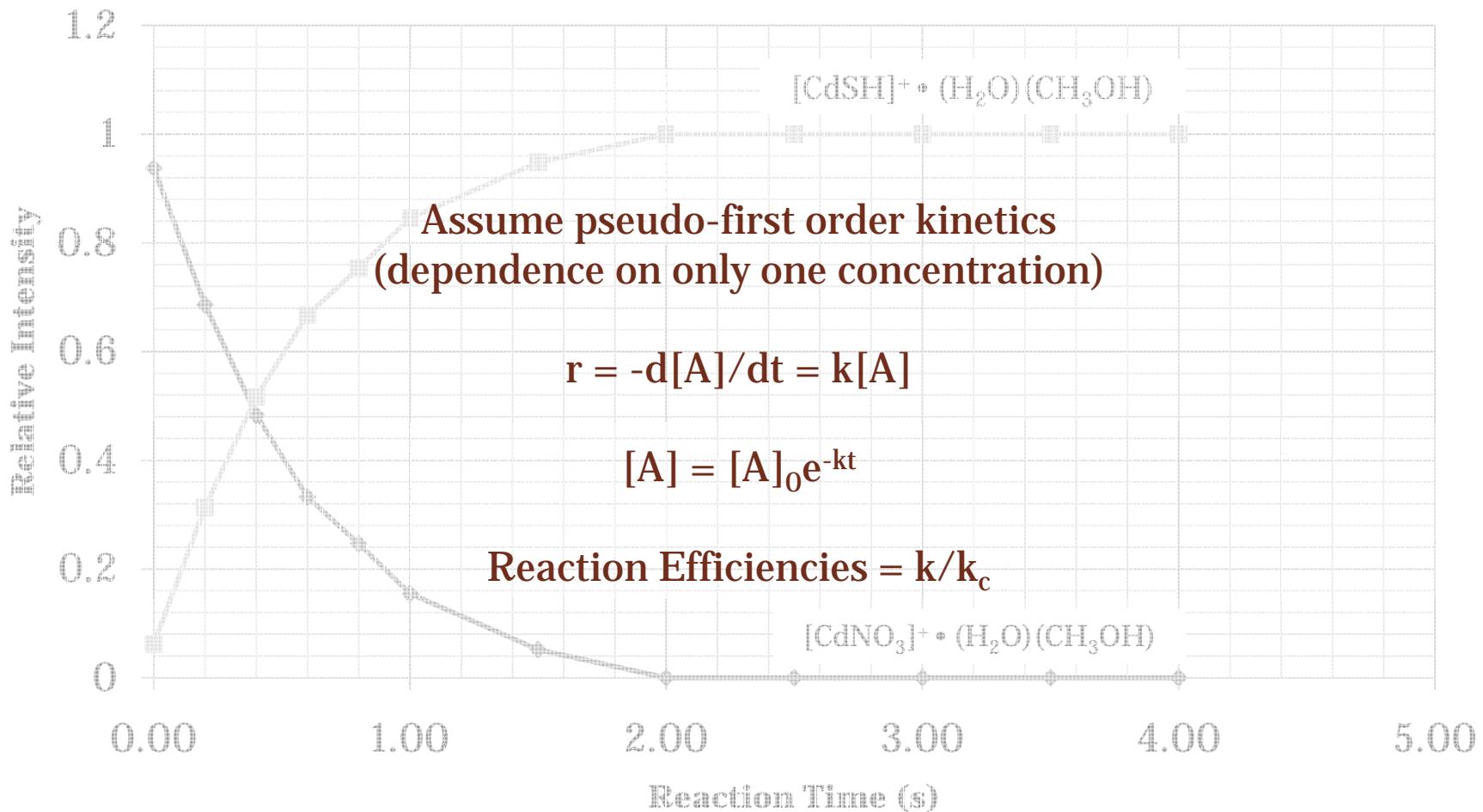
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Anions

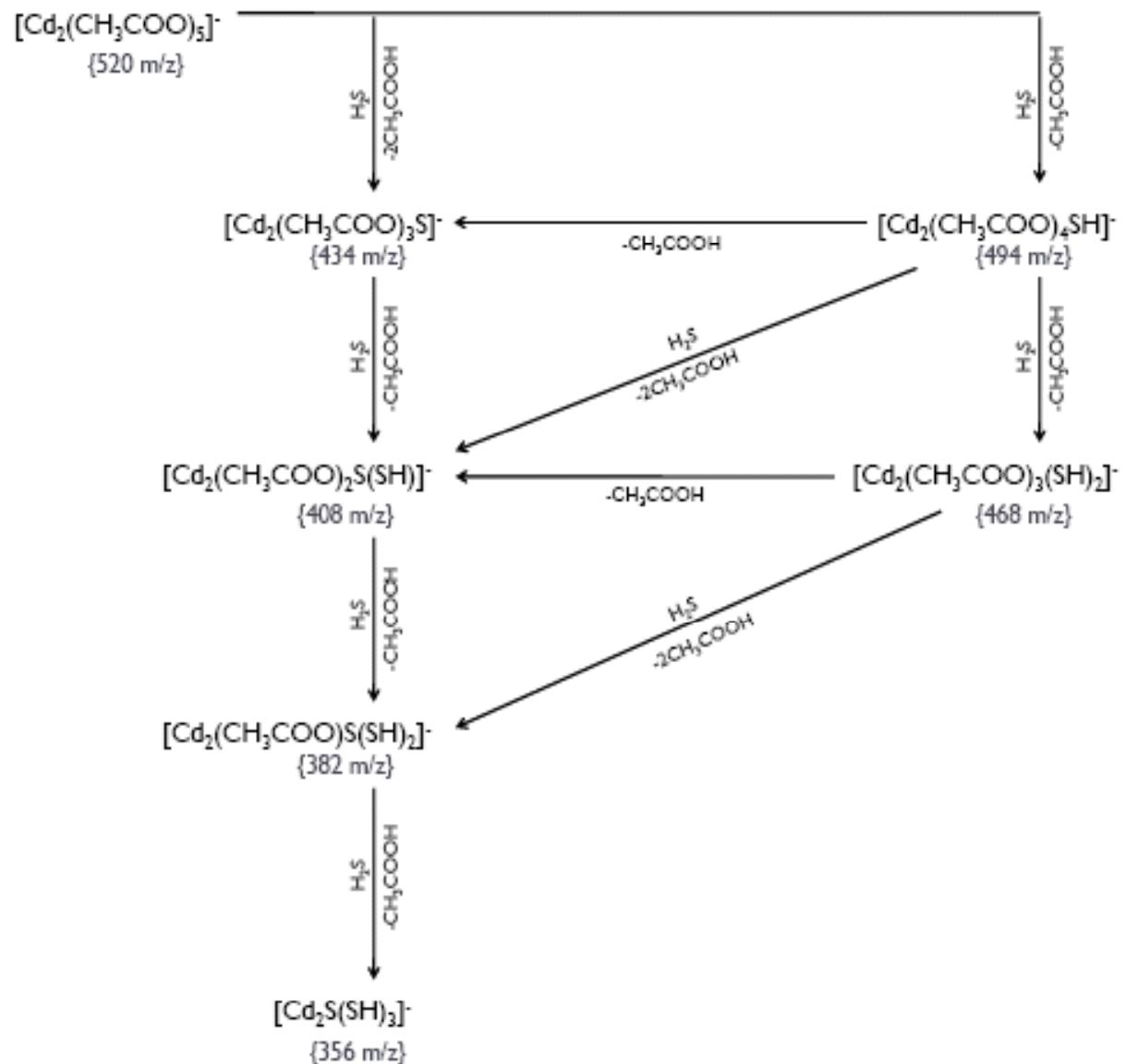
0.5 mM Cd(CH₃COO)₂ • 4(H₂O)

0.5 mM Cd(NO₃)₂ • 4(H₂O)

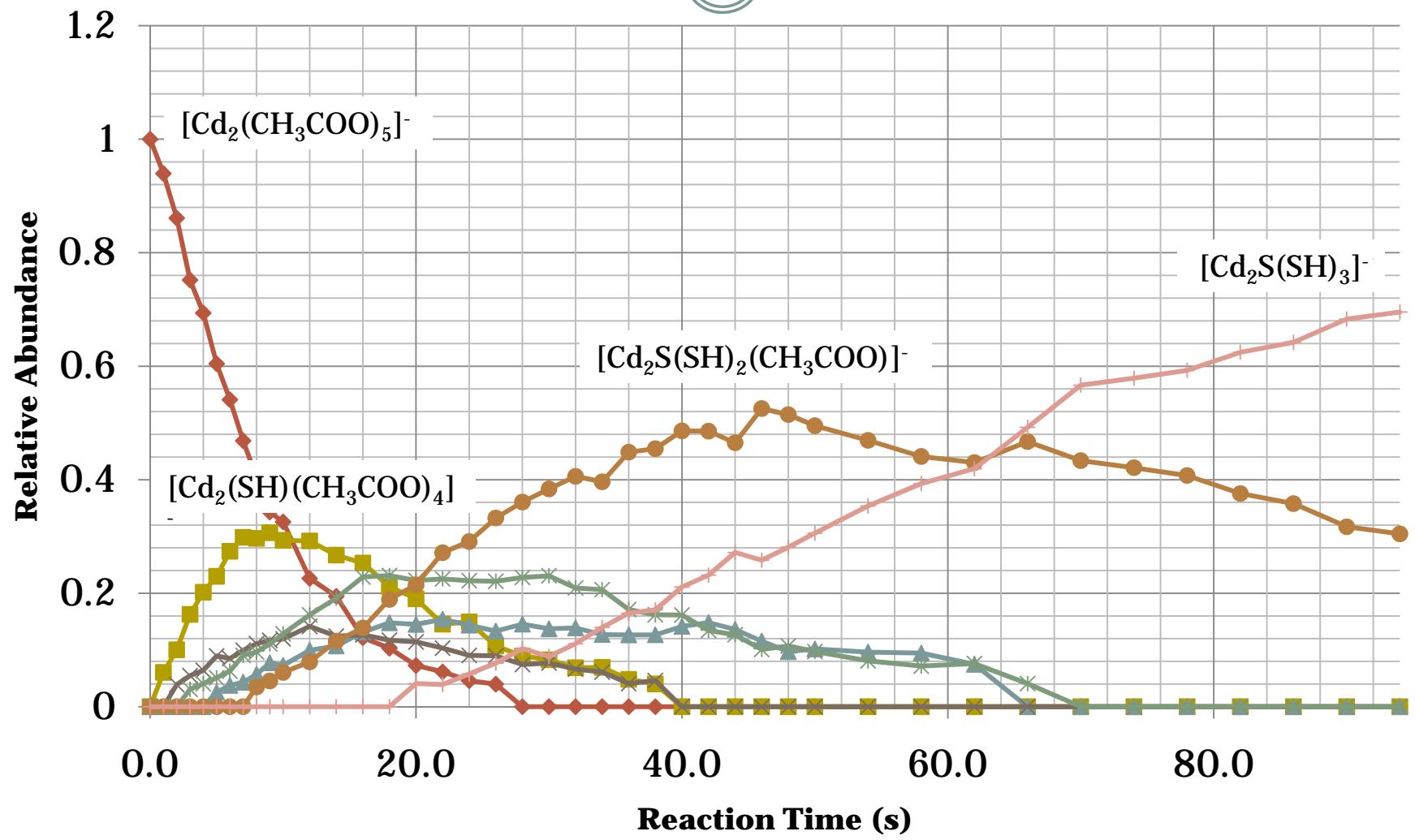
0.5 mM CdCl₂ • 4(H₂O)

1:1 water/MeOH

9x10⁻⁹ torr H₂S(g)



Reaction Kinetics



Significance

Making Clusters

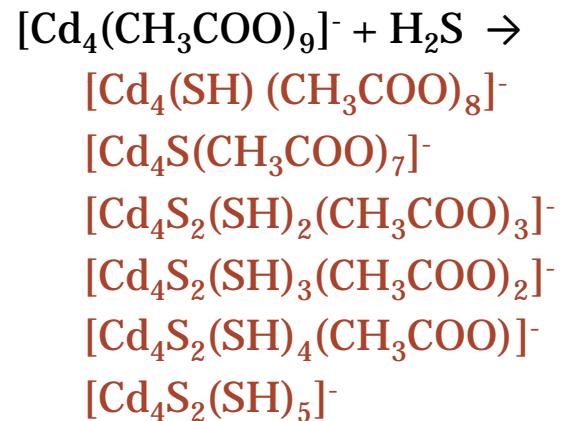
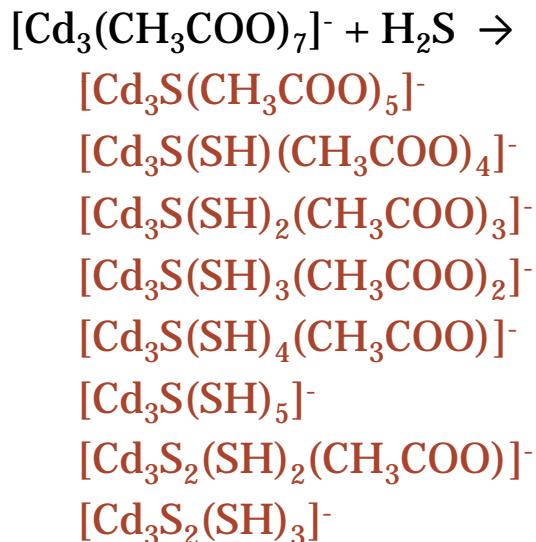
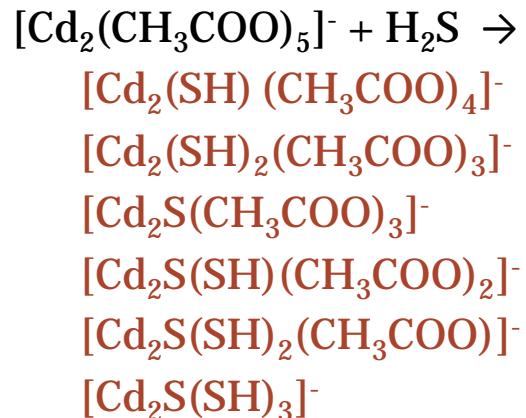
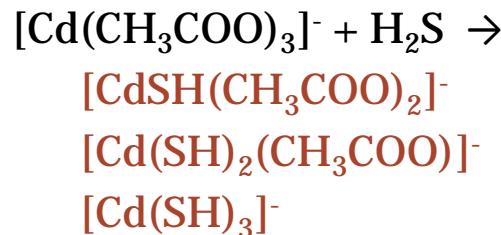
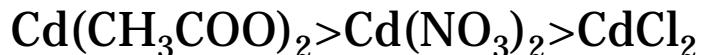
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Gas Phase Results

H_2S reactions with solvated CdNO_3^+ cations show solvent molecules to be spectators.

Reactivity varies with cadmium salt anion.

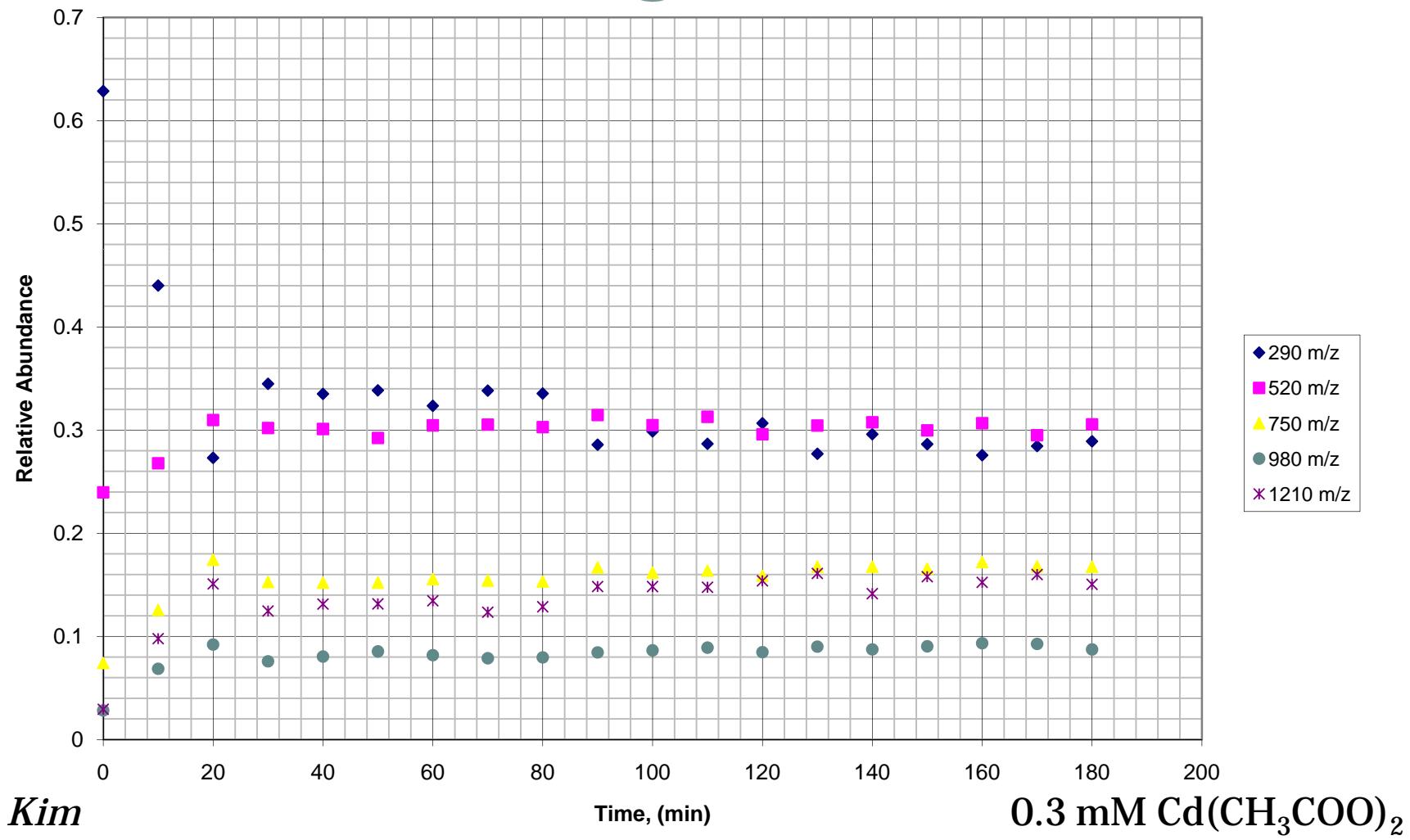


The Next Step



- High Pressure
 - Cd(NO₃)₂
 - CdCl₂
- CID Experiments
- Other Metals
 - Zn
 - Fe
 - Cu

Cluster Growth vs. Time



Una Kim

$0.3\text{ mM Cd}(\text{CH}_3\text{COO})_2$

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