Map it out. Use the ester bound carbon as the starting point. We can see that there is a bond that will be formed between C-5 and C-3

PhS 
$$\frac{5}{4}$$
 Br  $\frac{5}{4}$   $\frac{3}{2}$  MeO<sub>2</sub>C  $\frac{1}{4}$  CO<sub>2</sub>Me MeO<sub>2</sub>C  $\frac{1}{4}$  CO<sub>2</sub>Me

Start the problem. Use AIBN to generate a tin radical (see radical handout 1), then form the bond using a multiple bond addition reaction.

Map out the carbon atoms that you have not yet connected. It becomes apparent that C-6 needs to be connected to C-5. Use a multiple bond addition to accomplish it.

Finish the problem with a fragmentation reaction (the reverse of multiple bond addition). Then abstract a hydrogen to regenerate  $Bu_3Sn^{\bullet}$