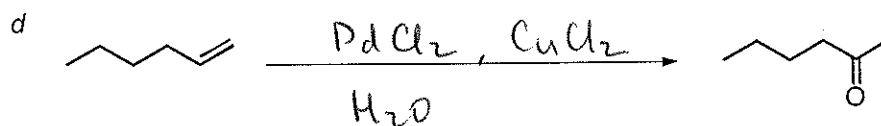
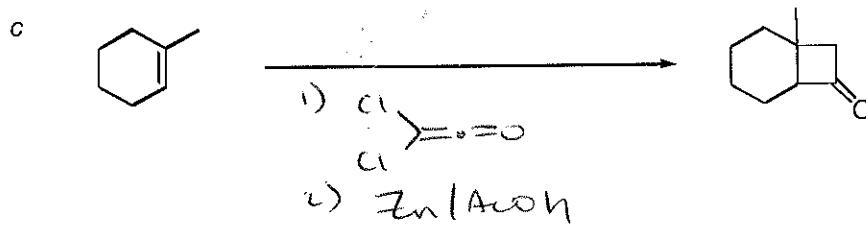
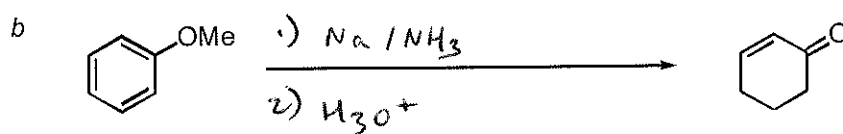
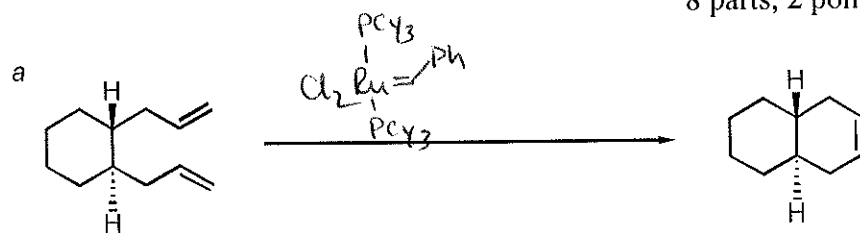


Chem 634  
Exam 2  
December 16, 2005  
3 hours  
Prof. Fox

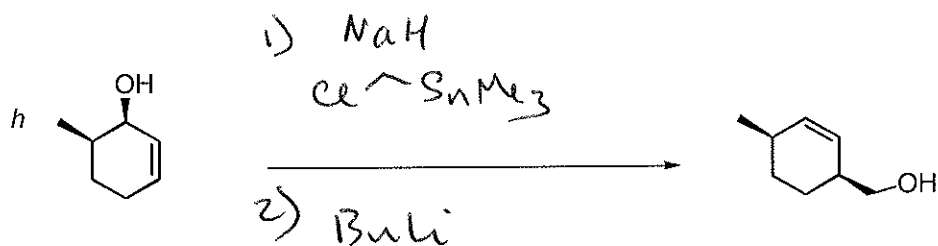
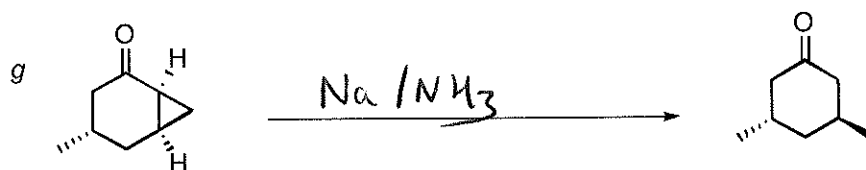
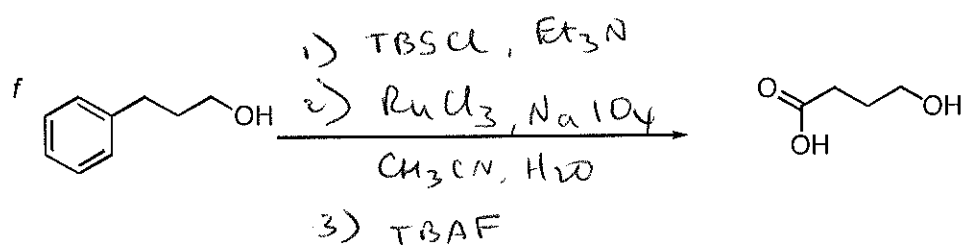
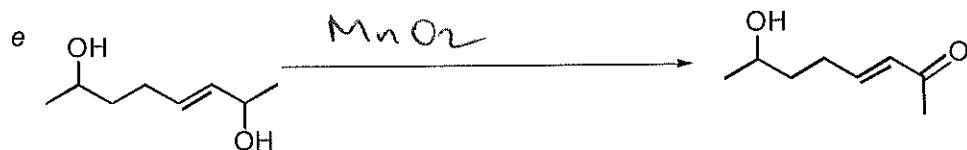
Your Name KEY

1. Provide reagents for the following transformation. More than one step may be required.  
Do not write mechanisms.

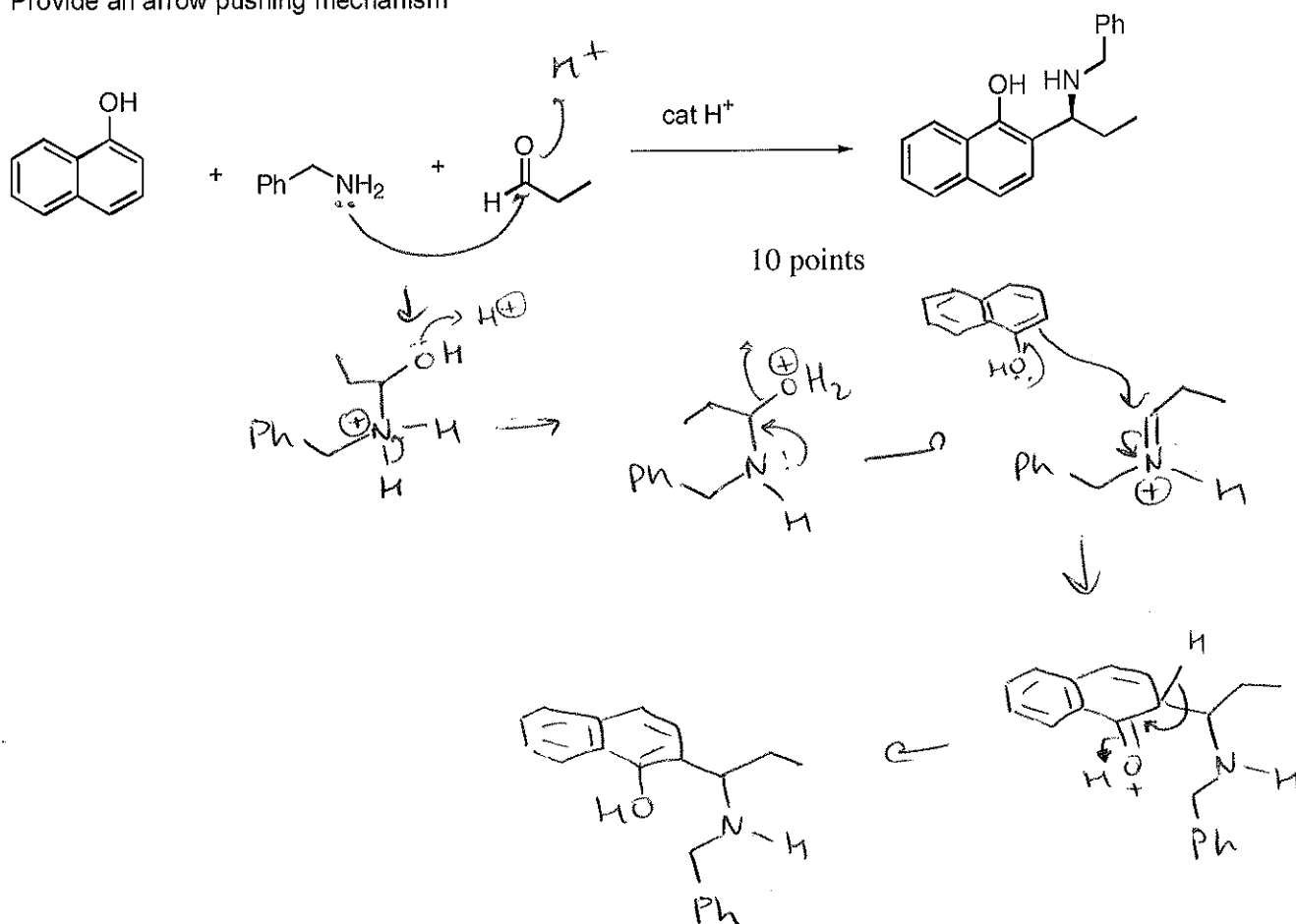
8 parts, 2 points each



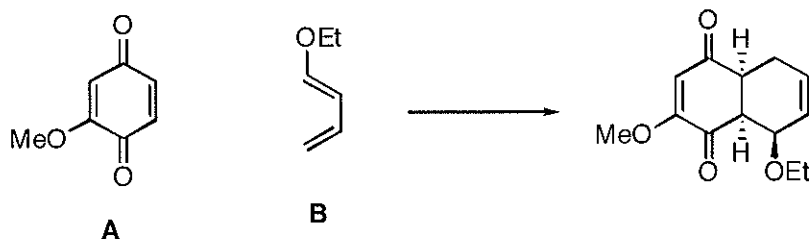
Question 1 continued



2. Provide an arrow pushing mechanism

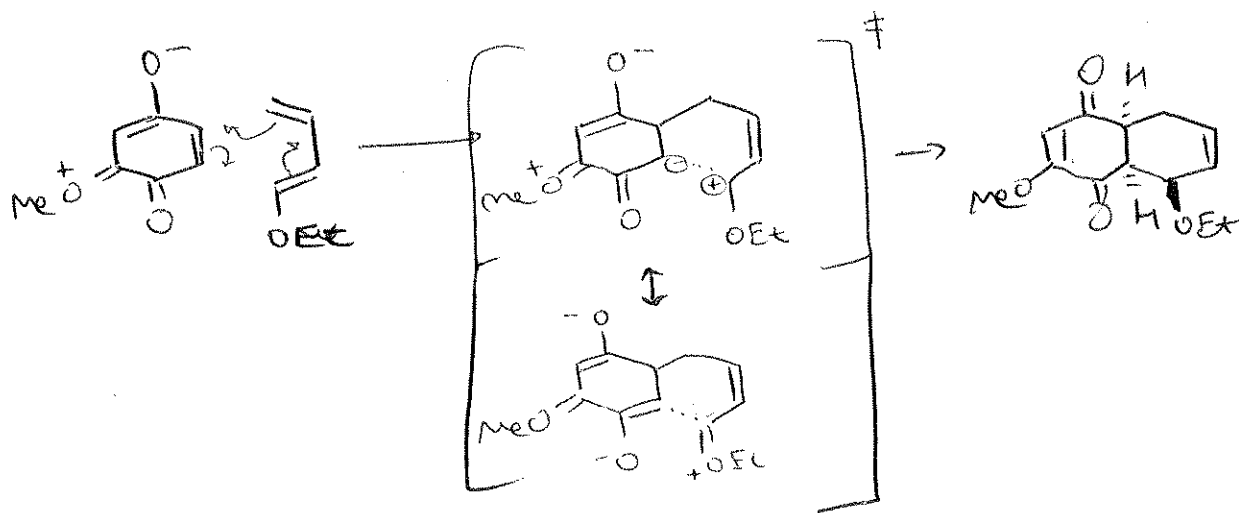
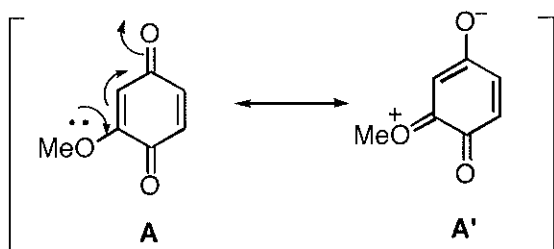


3(a) The reaction of diene **A** with dienophile **B** is regioselective. Explain.



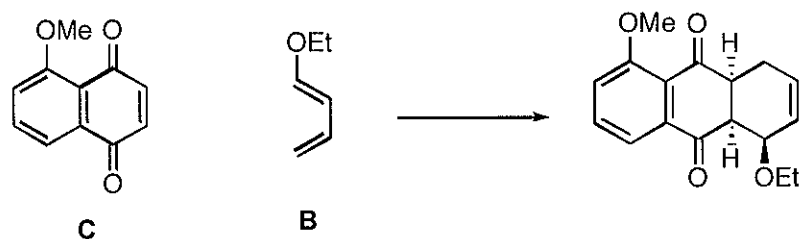
6 points

When preparing your answer, consider that the methoxy 'deactivates' the top carbonyl through resonance as shown below. You may assume that the resonance form **A'** is the dominating factor that guides the regioselectivity.

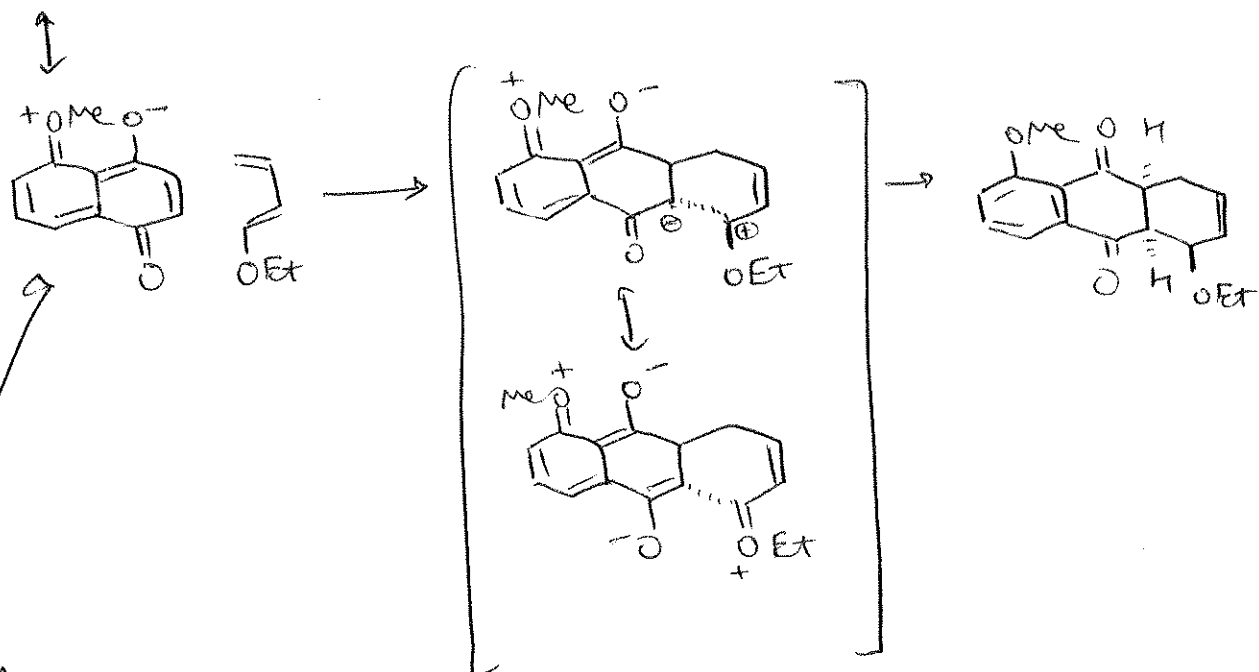


The concerted, asynchronous transition state has the "flavor" of a stepwise process with buildup of charge. The favored TS places <sup>PARTIAL</sup> positive charge next to the ethoxy, and partial negative charge next to the carbonyl.

3(b) The reaction of diene **C** with dienophile **B** is also regioselective. Explain.

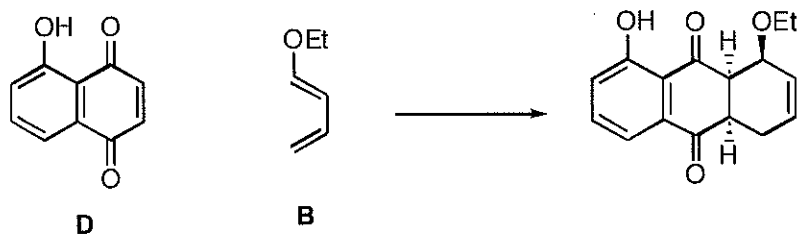


6 points



Asynchronous,  
Concerted TS Dominated by  
This resonance structure

3(c) The reaction of diene **D** with dienophile **B** displays opposite regioselectivity as compared to **C**. Explain.

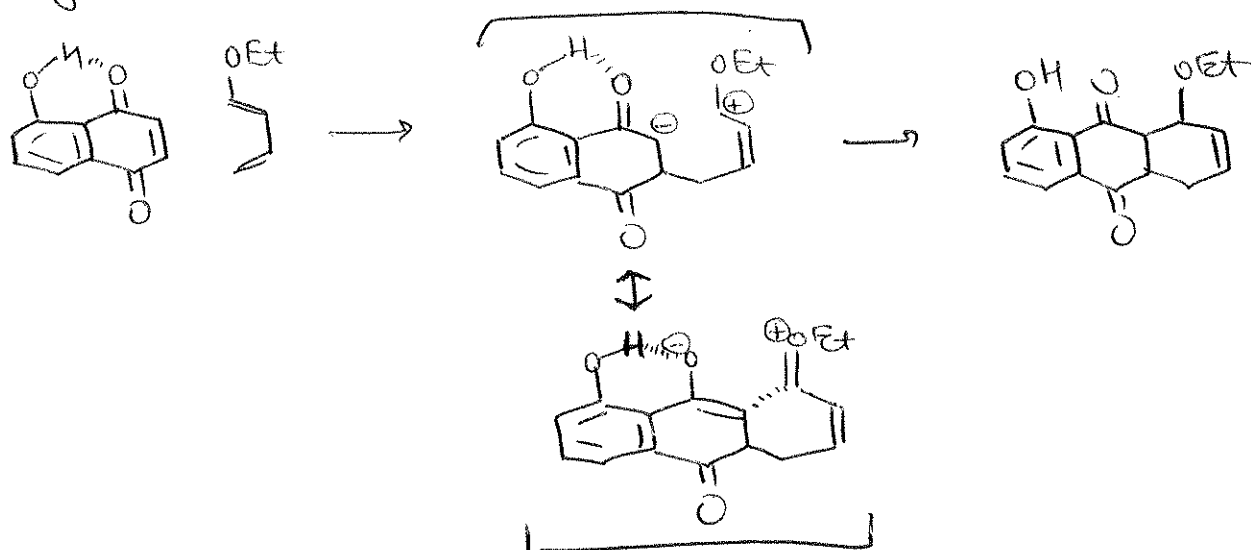


10 points

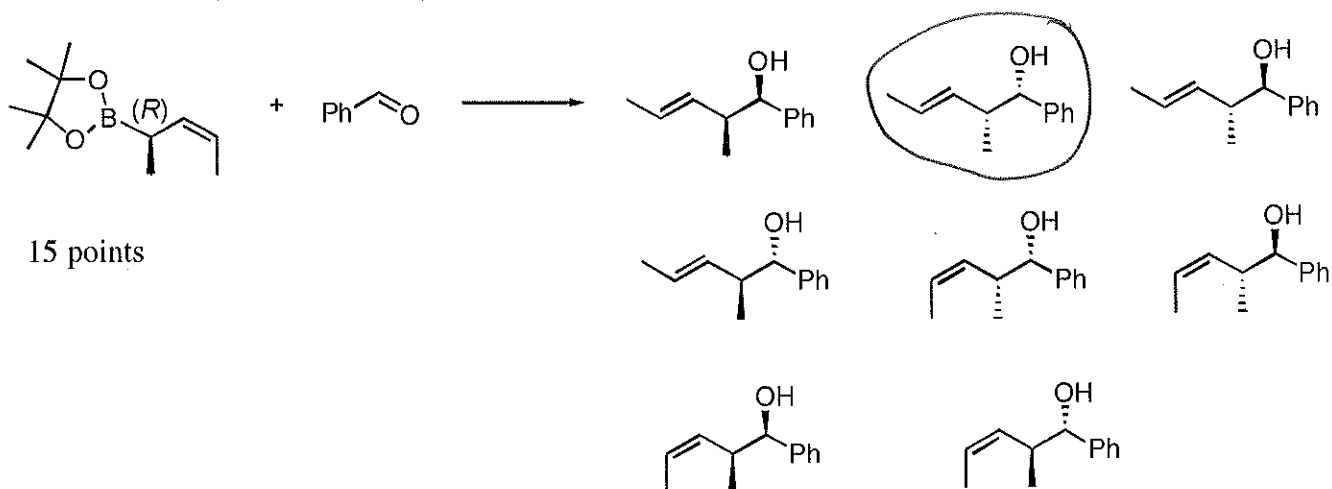
Hint 1: disregard what you wrote in parts (a) and (b) of this question. It is overwhelmed by another effect.

Hint 2: the phenolic hydrogen of **D** is acidic. Recall the effect that Lewis acids have on Diels Alder reactions

Hydroxyl acts as a "LOCAL" Bronsted Acid

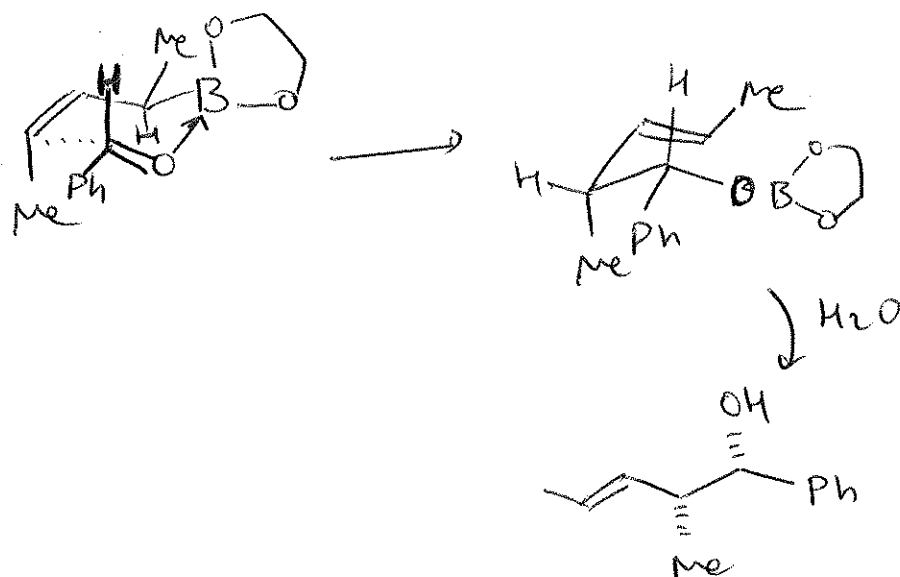


4. Circle the correct product. Explain your choice with a carefully drawn analysis of the transition state.



15 points

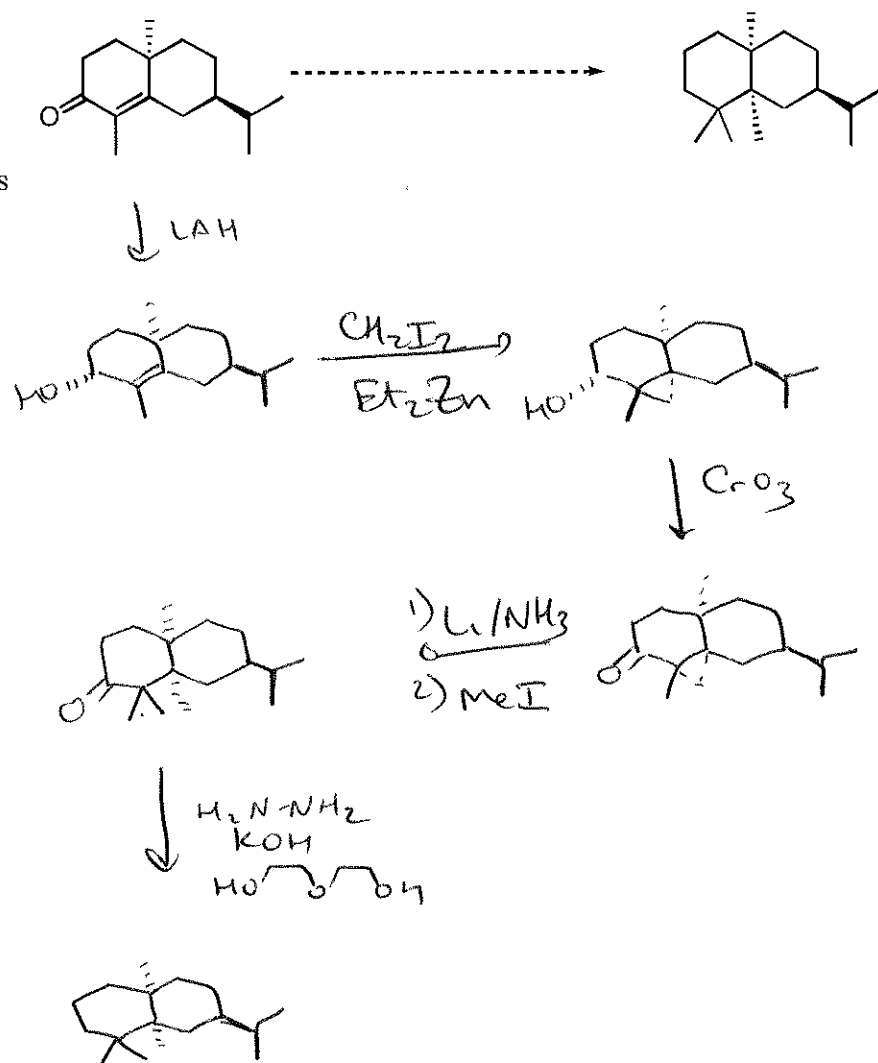
CHAIR TS



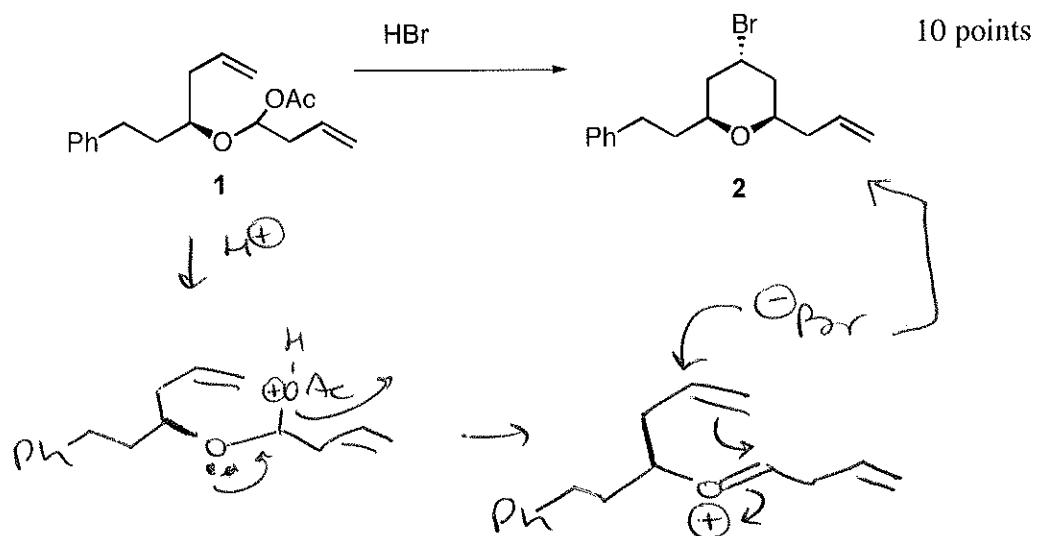


5. Propose a multistep synthesis

15 points

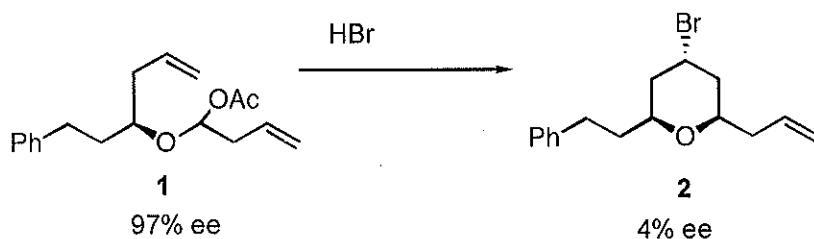


6(a) Propose an arrow pushing mechanism



6(b) When the reaction in 6(a) is carried out with enantiomerically enriched **1**, the product **2** that is obtained is nearly racemic. Propose an explanation

12 points



RATE OF  
RACEMIZATION VIA Claisen Rearrangement is  
COMPETITIVE WITH ATTACK OF BROMIDE

