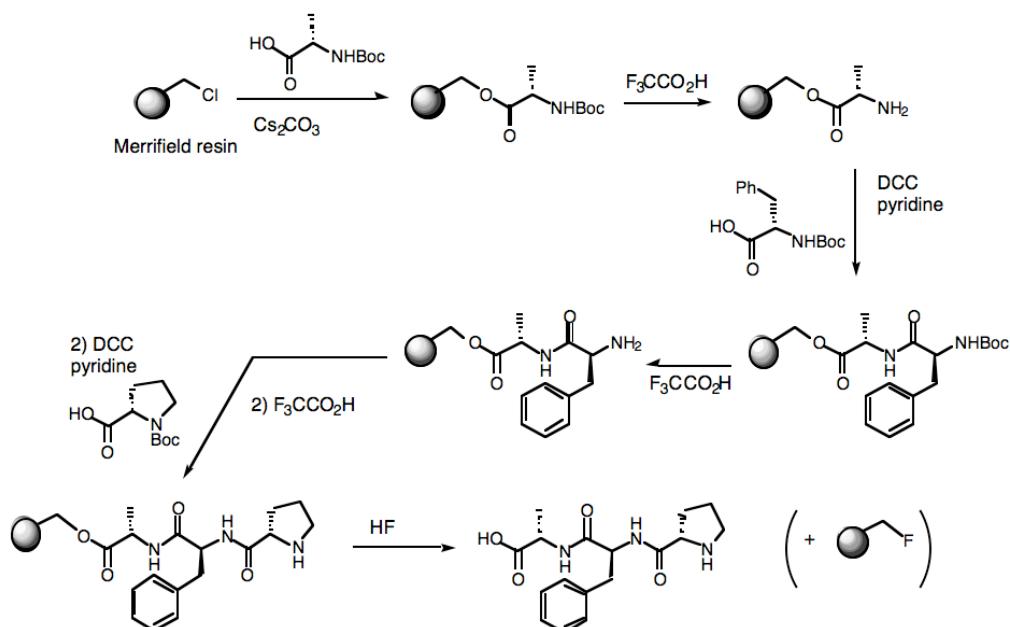
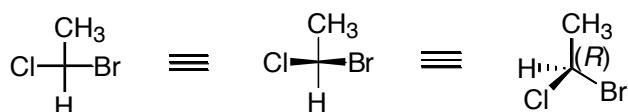


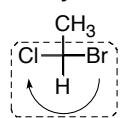
Solid Phase Peptide Synthesis



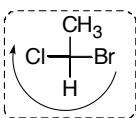
Fischer Projections revisited



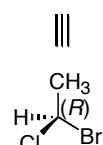
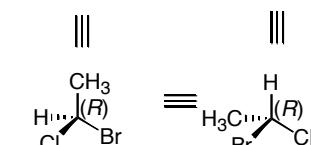
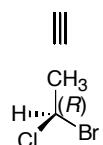
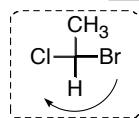
Remember that 3 group rotations are always allowed



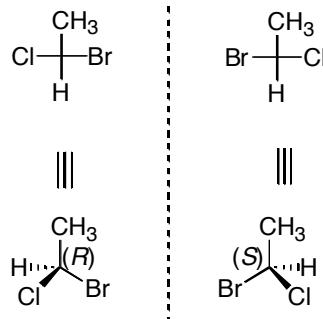
180° rotations of entire molecule are allowed



90° rotations of entire molecule are NOT allowed

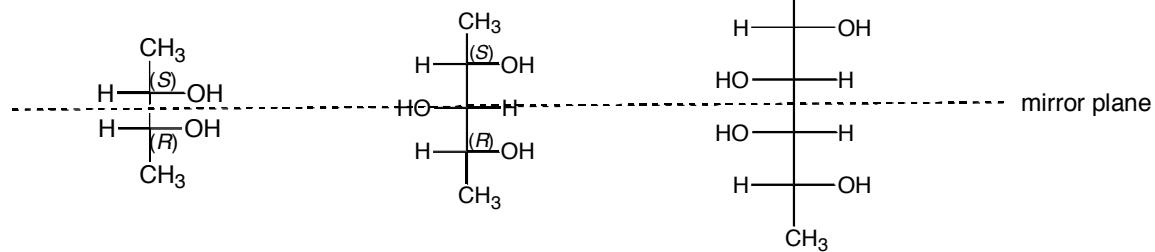


mirror images are enantiomers

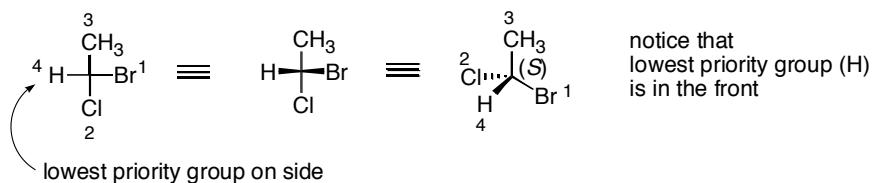
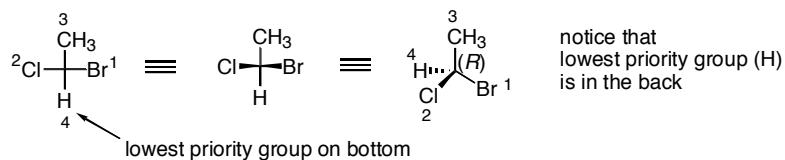


identification of meso compounds by a horizontal mirror plane

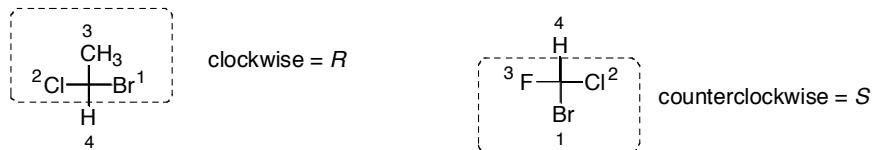
MESO compounds



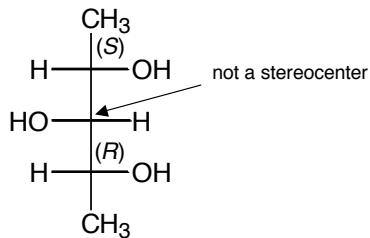
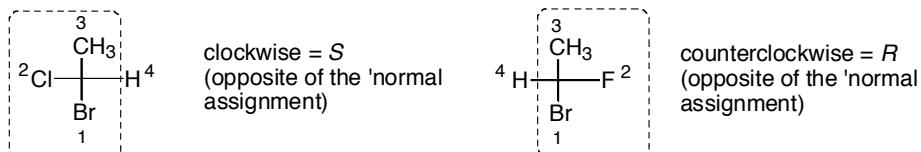
a Helpful way to assign stereochemistry



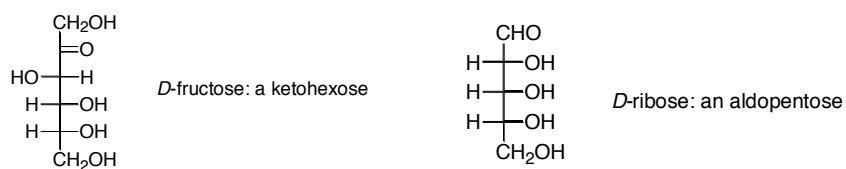
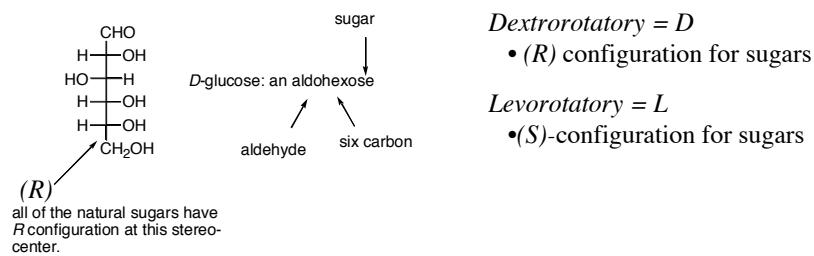
If the lowest priority group is on the top or bottom: assign 'as normal'



BUT If the lowest priority group is on the SIDE: make the OPPOSITE ASSIGNMENT

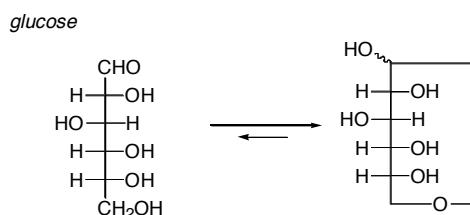
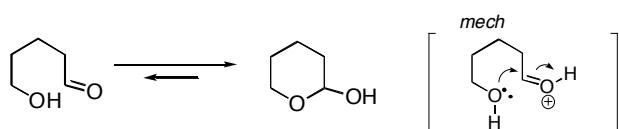


Carbohydrates: $C_nH_{2n}O$



Open chain sugars can equilibrate with cyclic hemiacetals

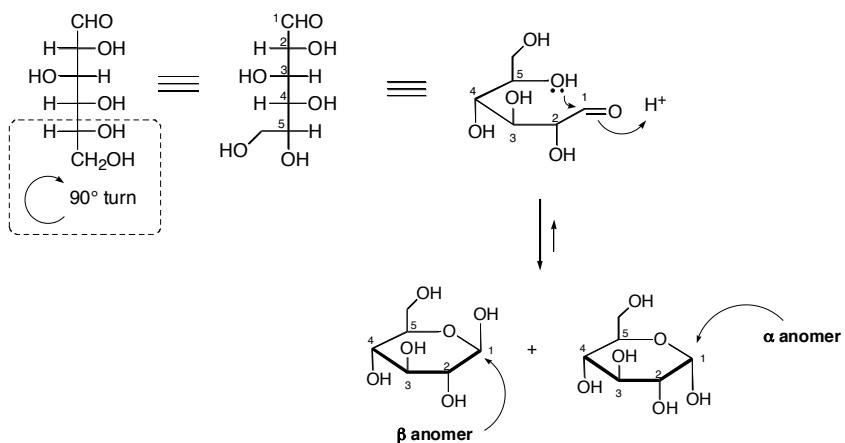
Cyclic hemiacetal formation



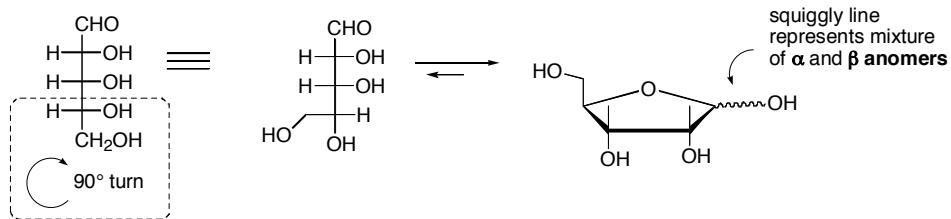
OK, so this is how they drew it back in the 1800's

Haworth Projections

For glucose



For ribose

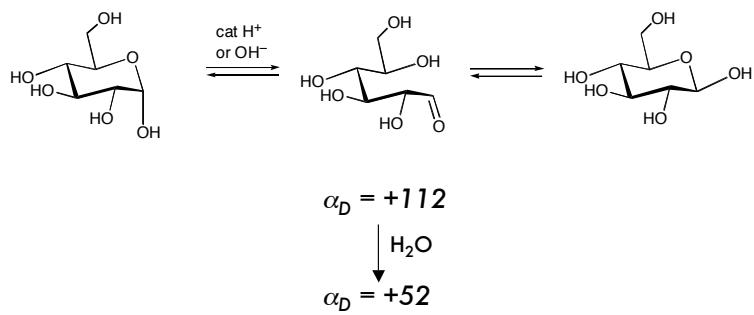


Converting Haworth Projections to Chair Projections

For glucose

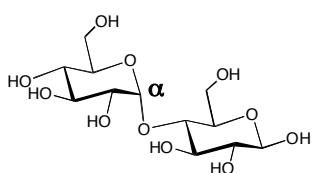


Mutarotation of glucose

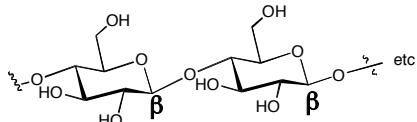


Oligosaccharides and Polysaccharides

a disaccharide



maltose (beer): a glucose dimer

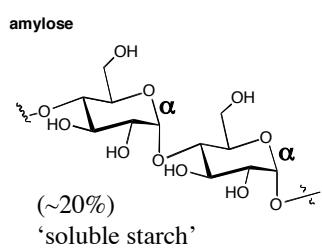


cellulose : a glucose polymer

highly linear structure: very effective
for intrastrand H-bonding

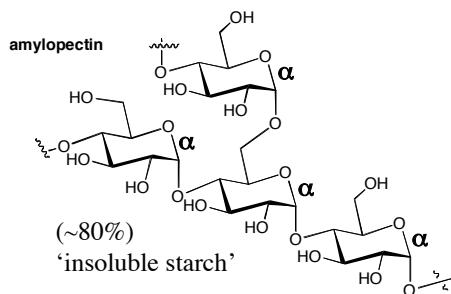
insoluble: cell walls

starch: two components



(~20%)

'soluble starch'



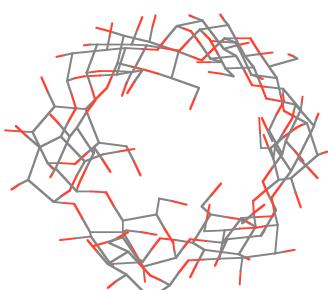
(~80%)

'insoluble starch'

Oligosaccharides and Polysaccharides

Amylose is highly helical

end view



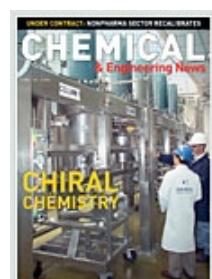
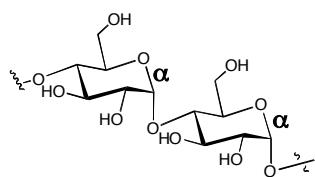
1-Naphthyl-2-butanol



Column = (S,S)-ULMO
25 cm x 4.6 mm
Mobile Phase = (95/5) Heptane/IPA
Flow Rate = 1.0 mL/min
Detection = UV 215 nm
Run Time = 6 min
 $K_1 = 0.80$
 $\alpha = 1.35$
reference 48

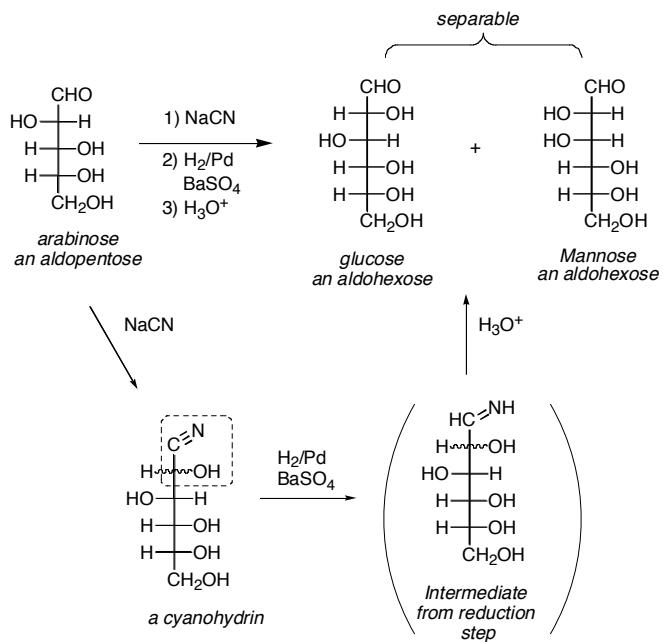


amylose



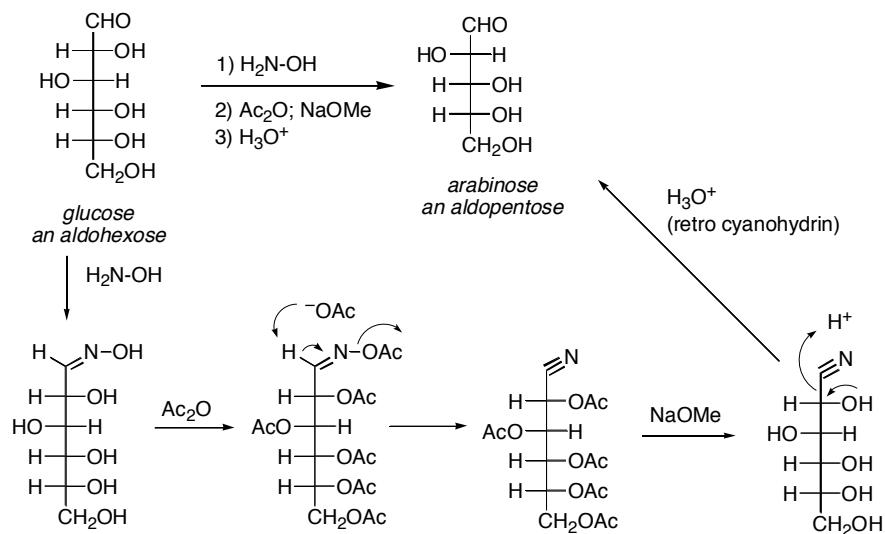
Some reactions of sugars

Kilian Fischer Synthesis

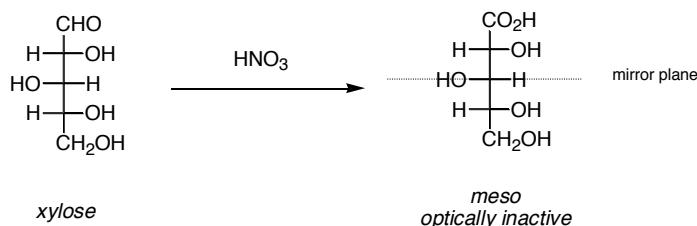
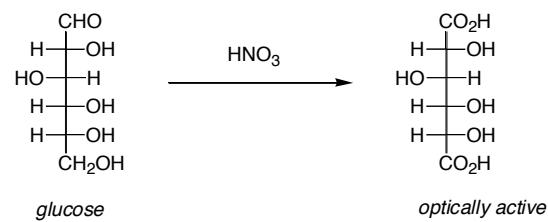


Some reactions of sugars

Wohl Degradation

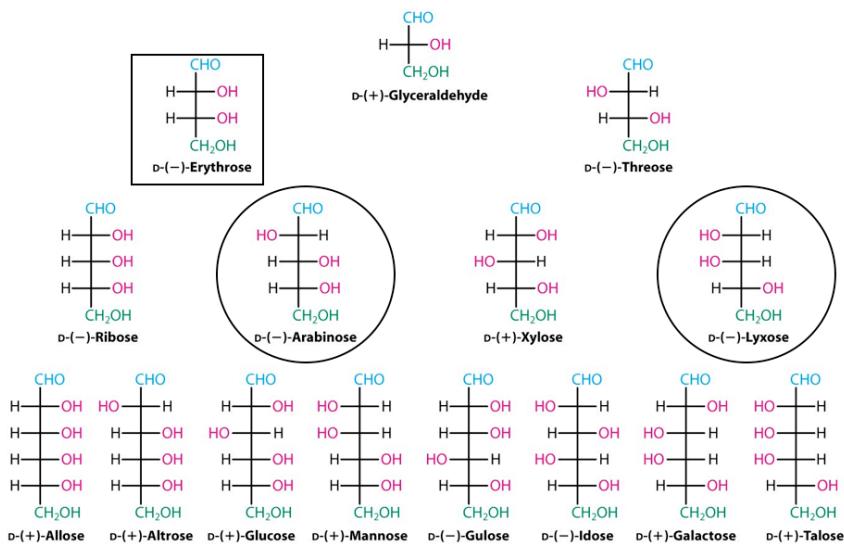


Oxidation of sugars



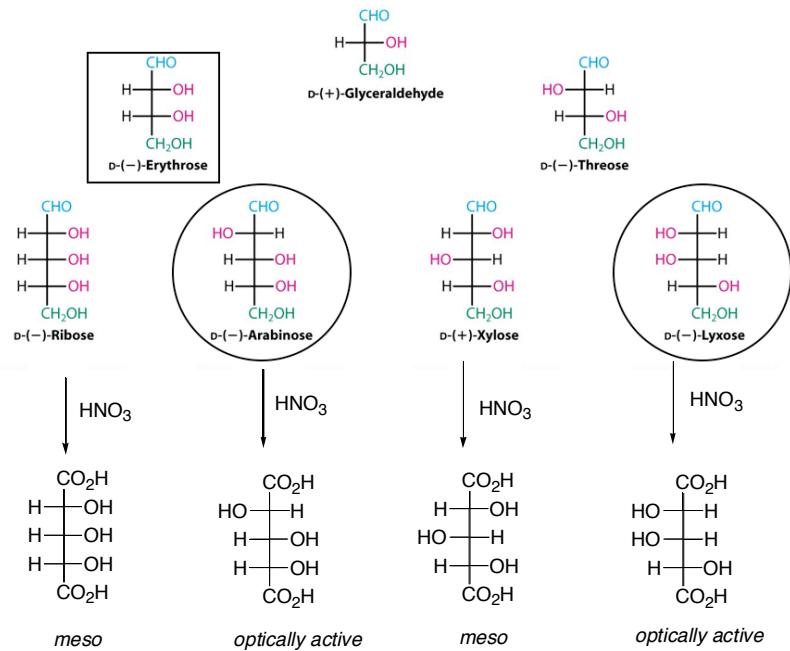
The Fischer Proof

1. An aldopentose is oxidized by HNO_3 to give an optically active diacid. Wohl degradation of that same aldopentose gives an aldotetrose. This aldotetrose provides an optically inactive diacid upon HNO_3 oxidation. Identify the aldopentose.



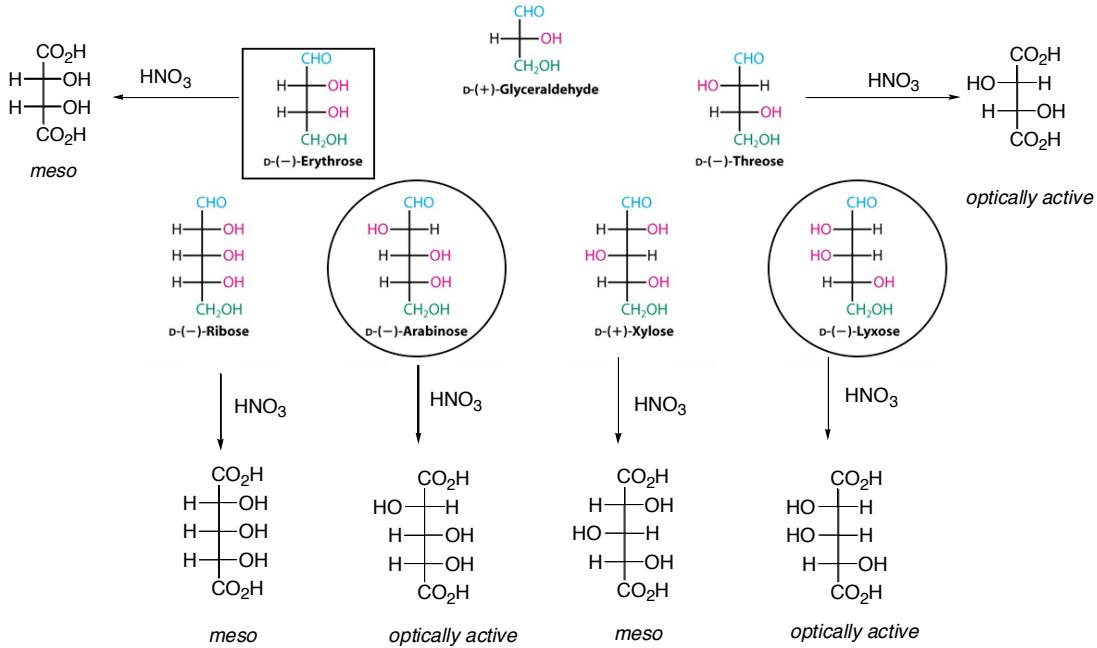
The Fischer Proof

1. An aldopentose is oxidized by HNO_3 to give an optically active diacid. Wohl degradation of that same aldopentose gives an aldotetrose. This aldotetrose provides an optically inactive diacid upon HNO_3 oxidation. Identify the aldopentose.



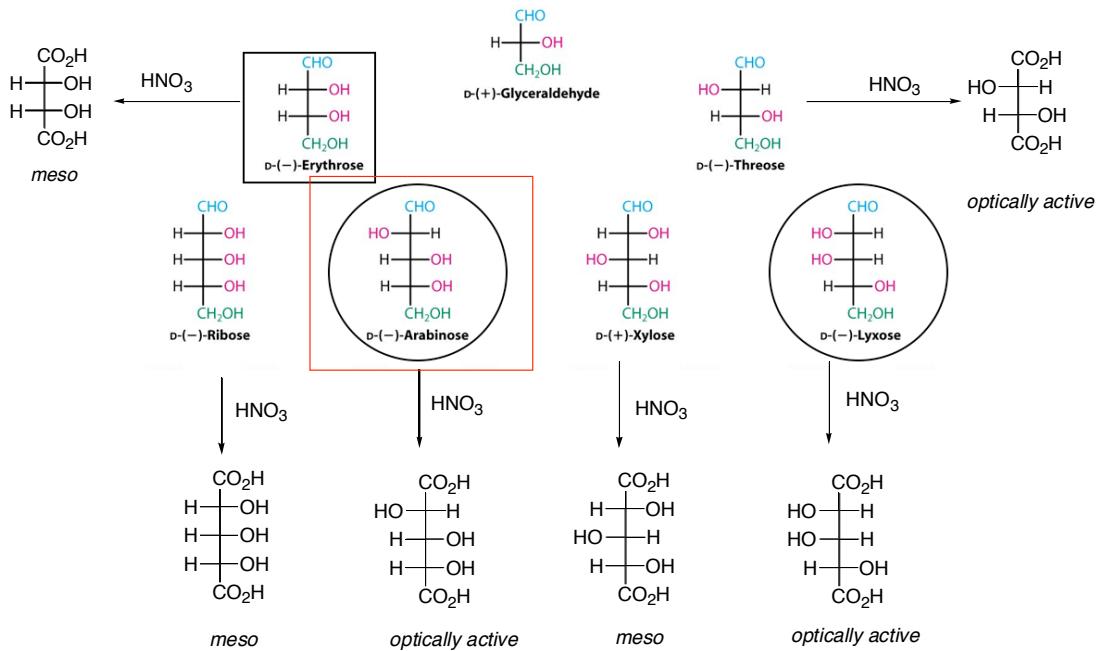
The Fischer Proof

1. An aldopentose is oxidized by HNO_3 to give an optically active diacid. Wohl degradation of that same aldopentose gives an aldotetrose. This aldotetrose provides an optically inactive diacid upon HNO_3 oxidation. Identify the aldopentose.



The Fischer Proof

1. An aldopentose is oxidized by HNO_3 to give an optically active diacid. Wohl degradation of that same aldopentose gives an aldotetrose. This aldotetrose provides an optically inactive diacid upon HNO_3 oxidation. Identify the aldopentose. **It is arabinose**



The Fischer Proof: Fischer used this type of analysis for all of the Sugars, starting from a good guess for glyceraldehyde

1. An aldopentose is oxidized by HNO_3 to give an optically active diacid. Wohl degradation of that same aldopentose gives an aldotetrose. This aldotetrose provides an optically inactive diacid upon HNO_3 oxidation. Identify the aldopentose. **It is arabinose**

