Chem 332 Exam 1 March 3, 2006 Prof. Fox 50 minutes 100 points

Show your work in detail

WRITE YOUR NAME ON EVERY PAGE

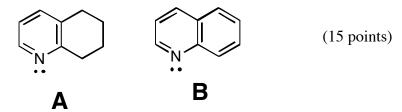
NAME	

1. Provide reagents for the following transformations (5 pts each)

$$Br$$
 NH_2

2a Circle the product below that is the stronger amine base. Provide a detailed but brief explanation to support your answer. Use chemical structures to support your answer. (15 points)

2b Consider compounds **A** and **B** below



An overly ambitious student might predict compound \mathbf{B} to be less basic than \mathbf{A} , because the lone pair of \mathbf{B} is conjugated with the neighboring aromatic ring as shown below.

$$\begin{bmatrix} \\ \\ \\ \\ \\ \\ \\ \\ \\ \end{bmatrix}$$

However, the analysis above is **incorrect:** the pKa's of the conjugate acids of $\bf A$ and $\bf B$ are nearly identical. Explain the flaw in the rationale above.

3 Provide a detailed arrow pushing mechanism. Show the mechanism of all steps.

4 Provide a synthesis of **A** using any materials that contain **3 carbons or less.** Reagents that do not become incorporated into the product (e.g. nBuLi, PPh3) may be employed

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NAME

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