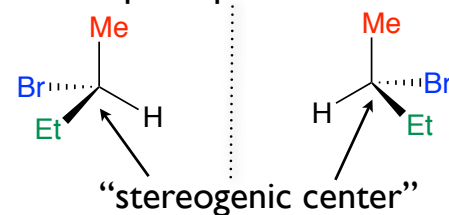


## chirality

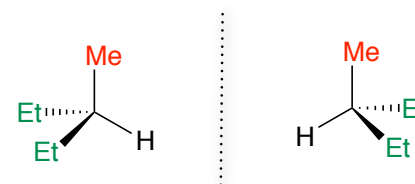
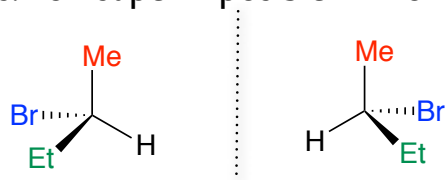


enantiomers: non superimposable mirror images



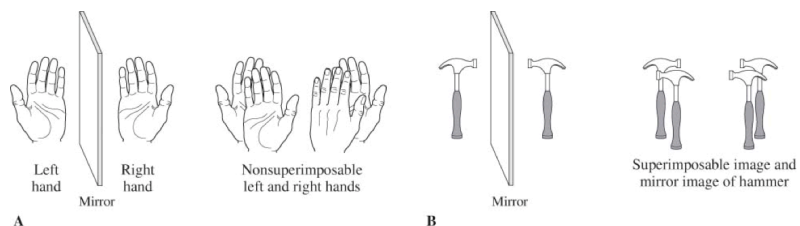
formalism:  
wedge (forward)  
dashed (back)

Not all mirror images are chiral. These are the same (not chiral)

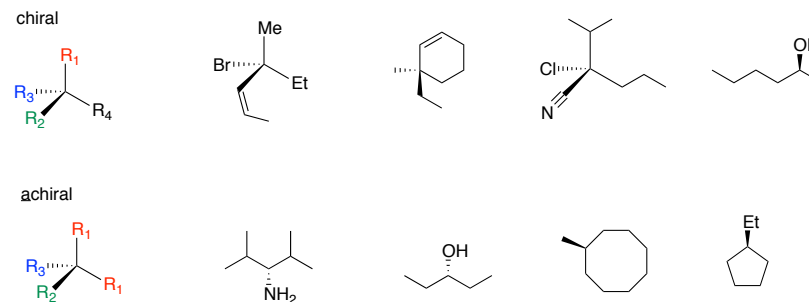


enantiomers: non superimposable mirror images

Another type of stereoisomerism is called [mirror-image stereoisomerism](#). Mirror-image related stereoisomers are said to be left-handed and right-handed and occur when a molecule and its mirror image are non-superimposable.



How to tell if molecules are chiral? Clue: many (but not all\*) chiral carbons have 4 different things attached



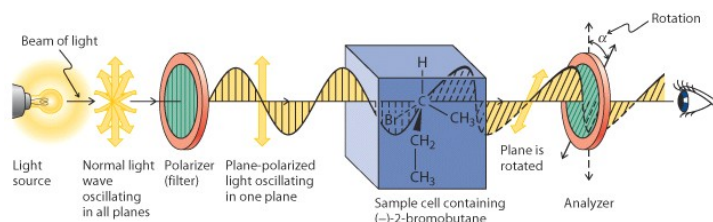
enantiomers  
have same mp, bp, NMR, IR, etc

\* exceptions to be discussed

When plane-polarized light interacts with a chiral molecule, the plane of polarization of the light is rotated to the left or right.

This effect is called optical rotation and the molecule is said to be optically active.

Optical activity is measured using a polarimeter. This device contains a light source, a polarizer to produce the plane-polarized light, a sample cell, and an analyzer to determine the amount of rotation.



The measured rotation (in degrees) is the **observed optical rotation**,  $\alpha$ , of the sample.

Specific rotation is a physical constant for a substance, as is melting point, boiling point, density, etc.

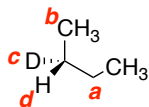
$\begin{array}{c} \text{CH}_2\text{CH}_3 \\   \\ \text{H}-\text{C}-\text{CH}_3 \\   \\ \text{Br} \end{array}$ (-)-2-Bromobutane	-23.1	$\begin{array}{c} \text{CH}_2\text{CH}_3 \\   \\ \text{H}_3\text{C}-\text{C}-\text{H} \\   \\ \text{Br} \end{array}$ (+)-2-Bromobutane	+23.1
$\begin{array}{c} \text{H} \\   \\ \text{H}_2\text{N}-\text{C}-\text{COOH} \\   \\ \text{CH}_3 \end{array}$ (+)-2-Aminopropanoic acid [(+)-Alanine]	+8.5	$\begin{array}{c} \text{H} \\   \\ \text{HOOC}-\text{C}-\text{OH} \\   \\ \text{CH}_3 \end{array}$ (-)-2-Hydroxypropanoic acid [(-)-Lactic acid]	-3.8

Note: Pure liquid for the haloalkane; in aqueous solution for the acids.

## Cahn Ingold Prelog rules

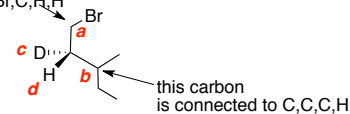
prioritize by:

- 1) Highest atomic #
- 2) then by isotope atomic weight
- 3) then by substituents, one atom at a time

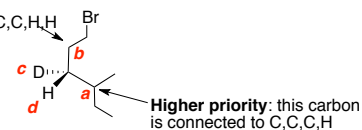


## 4) if atoms are equal, go to first point of difference

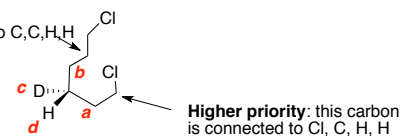
**Higher priority:** this carbon is connected to Br, C, H, H



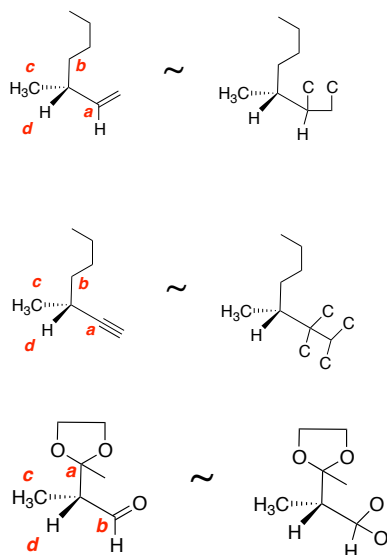
this carbon is connected to C, C, H, H



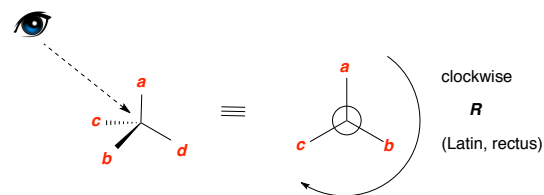
this carbon is connected to C, C, H, H



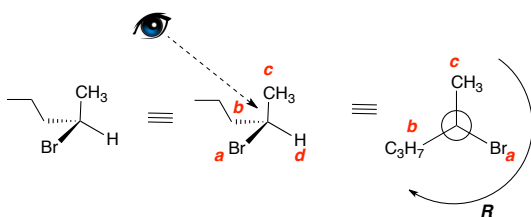
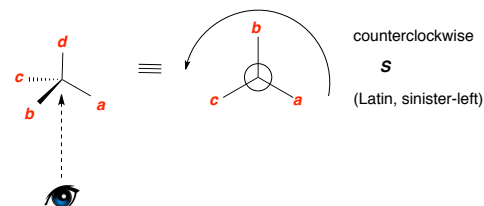
## 5) multiple bonds



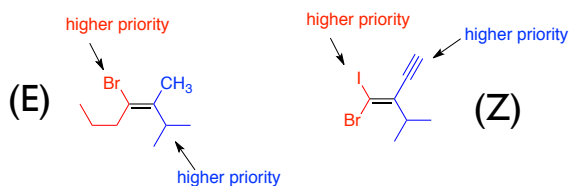
View the molecule with the lowest priority group in the back



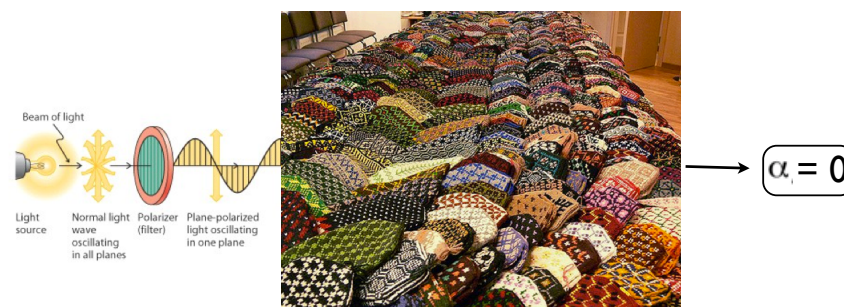
opposite enantiomer:



same rules for alkenes



Chiral molecules are not necessarily optically active

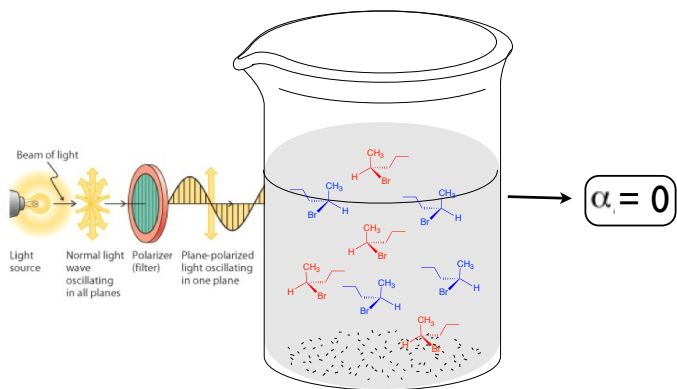


[Pile of Latvian Mittens](#) (50% Left, 50% Right)

The Latvians are knitting 4,500 mittens for the participants of the NATO summit there (2006)!

See here: [www.rigasummit.lv/en/?id=newsin&nid=115](http://www.rigasummit.lv/en/?id=newsin&nid=115)

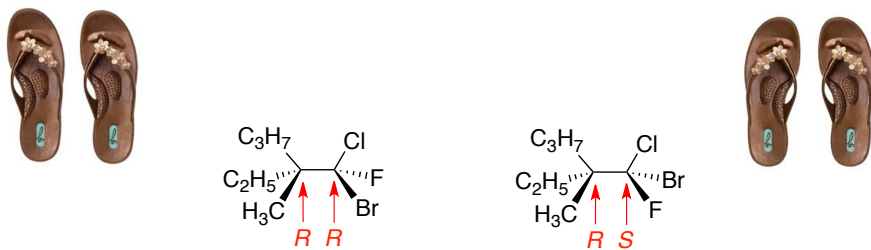
Racemates are not optically active



What happens if an enantiomer interacts with another element of chirality?

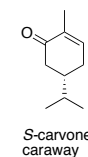
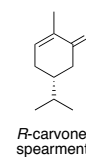
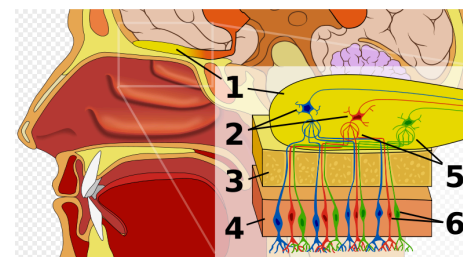


What happens if an enantiomer interacts with another element of chirality?



diastereomers  
have different mp, bp, NMR, IR, etc

diastereomeric pairs up the nose



## diastereomers

easy to see in ring compounds

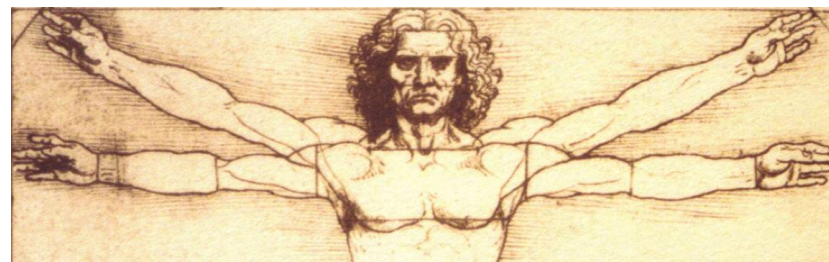


easy to see in acyclic compounds if we are consistent in our drawings



## meso compounds

Meso compounds contain an internal mirror plane (or center of symmetry) which divides the two halves of the molecule which are mirror images of each other. NOT OPTICALLY ACTIVE!



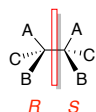
Vitruvian man = meso man. He has symmetrically substituted elements of chirality, equal and opposite

## meso compounds

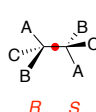
recognizing planes and centers of symmetry



plane of symmetry

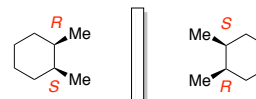


center of symmetry

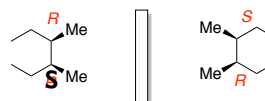


## meso compounds

easy to see in ring compounds

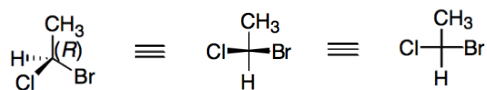


easy to see in acyclic compounds if we are consistent in our drawings

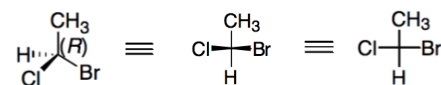


in general, look for R & S stereocenters in the same molecule attached to exactly the same thing

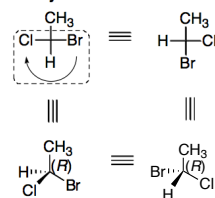
### Fischer Projections



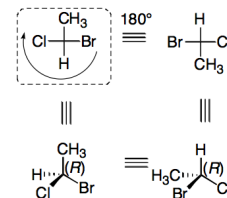
### Fischer Projections revisited



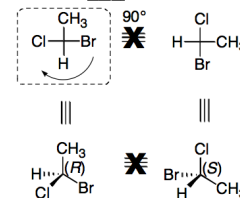
Remember that 3 group rotations are always allowed



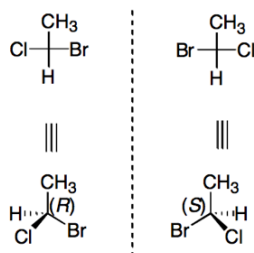
180° rotations of entire molecule are allowed



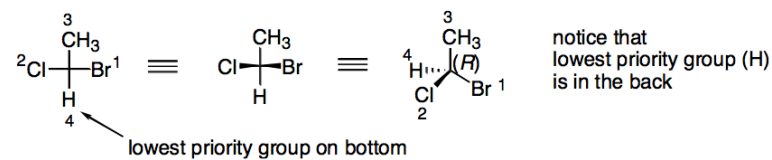
90° rotations of entire molecule are NOT allowed



mirror images are enantiomers



a Helpful way to assign stereochemistry



identification of meso compounds by a horizontal mirror plane

MESO compounds

