

## Fluid, Electrolyte, and Acid Base

Part 6 Acid Base Balance  
and Disorders

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## Regulation of Acid Base Balance

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## Body Buffering Systems

- Bicarbonate and carbonic acid-largest; in ECF
- $\text{NaH}_2\text{PO}_4$ -Phosphate is important buffer in RBCs and renal tubules
- Hemoglobin and oxyhemoglobin buffer systems
- Protein buffer systems

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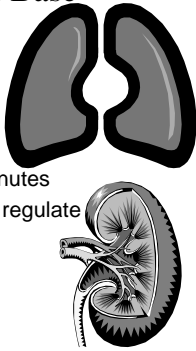
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## Homeostasis of Acid Base Balance

- Blood buffer system-can accept or donate  $H^+$ ; rapid
- Lungs-control ECF  $CO_2$ ;  
■ quick acting; effective in minutes
- Kidney-excrete excess  $H^+$ ; regulate bicarbonate reabsorption;  
slow acting-requires hours to days for effect



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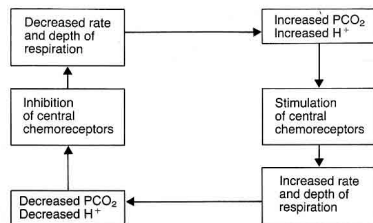
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## CNS Control of Respiration



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## Acidosis

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## Causes of Respiratory Acidosis

- Respiratory depression
- Inadequate chest expansion
- Airway obstruction
- Alveolar-capillary block




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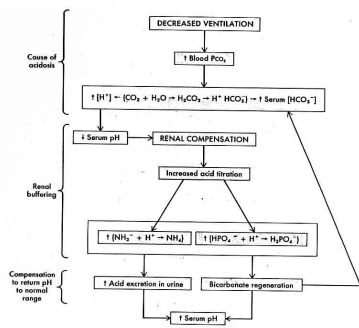
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## Respiratory Acidosis




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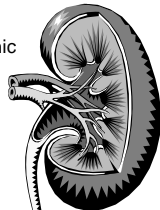
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## Metabolic Acidosis

- Definition: Systemic disorder characterized by a primary decrease in plasma bicarbonate
- Respiratory compensation starts immediately
- Etiology:
  - Increased fixed or noncarbonic acids
  - Renal failure
  - Loss of base bicarbonate




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## Metabolic Acidosis

- Definition: Primarily a decrease in plasma bicarbonate
- Causes:
  - overproduction of  $H^+$
  - underelimination of  $H^+$
  - underproduction of  $HCO_3^-$
  - Overerelimination of  $HCO_3^-$

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## Manifestations of Acidosis

- CNS: decreased activity-lethargy, confusion, stupor, coma
- Neuromuscular: hyporeflexia, weakness, flaccid paralysis
- Cardiac: delayed electrical conduction, hypotension, thready pulse
- Respiratory: Kussmaul's respirations, variable respirations

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## Alkalosis

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## Respiratory Alkalosis

- Pathology-Excess CO<sub>2</sub> loss
- Etiology
  - hyperventilation
  - central chemoreceptor stimulation
  - peripheral chemoreceptor stimulation
    - hypoxemia
    - asphyxiation
    - high altitudes
    - shock




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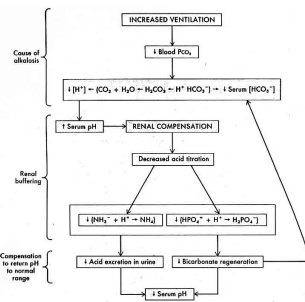
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## Respiratory Alkalosis




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## Metabolic Alkalosis

- **Definition:** plasma bicarbonate excess resulting in increased pH
- **Etiology**
  - **Increased base components**
    - Antacid ingestion
    - Blood transfusions
    - Sodium bicarbonate
    - TPN
    - Ringer's lactate excess
  - **Decreased acid components**
    - Prolonged vomiting, NG suctioning
    - Cushing's syndrome, hyperaldosteronism
    - Thiazide diuretics




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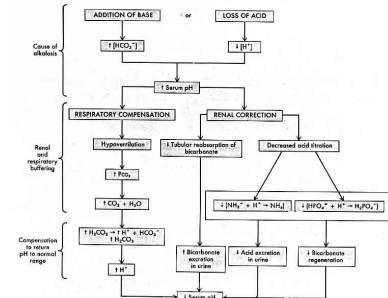
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## Metabolic Alkalosis



## Metabolic Alkalosis: Compensation

- Intracellular buffering- $H^+$  leaves cells to buffer
- Carotid chemoreceptors sense increased pH resulting in decreased ventilation
- Renal excretion of bicarbonate

## Common Manifestations of Alkalemia

- **CNS:** increased activity
- **Neuromuscular**
  - skeletal muscle weakness
  - Muscle cramping and twitching
  - hyperreflexia
- **Cardiac**
  - Tachycardia
  - Normo- to hypotensive blood pressure
- **Respiratory**