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Inaugural Fraunhofer – Delaware Technology Summit

***Inaugural Fraunhofer – Delaware
Technology Summit***

Energy and Life Sciences – Solutions for Sustainability

*University of Delaware
Clayton Hall
March 5/6, 2013*


 **Fraunhofer**  

Inaugural Fraunhofer – Delaware Technology Summit

SESSION B.3: SUSTAINABLE ENERGY CONCEPTS

Plenary Talk
*“Infrastructure-Ready Biomass-Derived Oils:
Incentives, Options and Process Models”*

- Michael Klein
Dan Rich Chair of Energy,
Chemical and Biomolecular Engineering
UD Energy Institute (UDEI)
University of Delaware





Infrastructure-Ready Biomass-Derived Oils: Incentives, Options and Process Models

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Energy Institute
University of Delaware
Newark, DE 19716

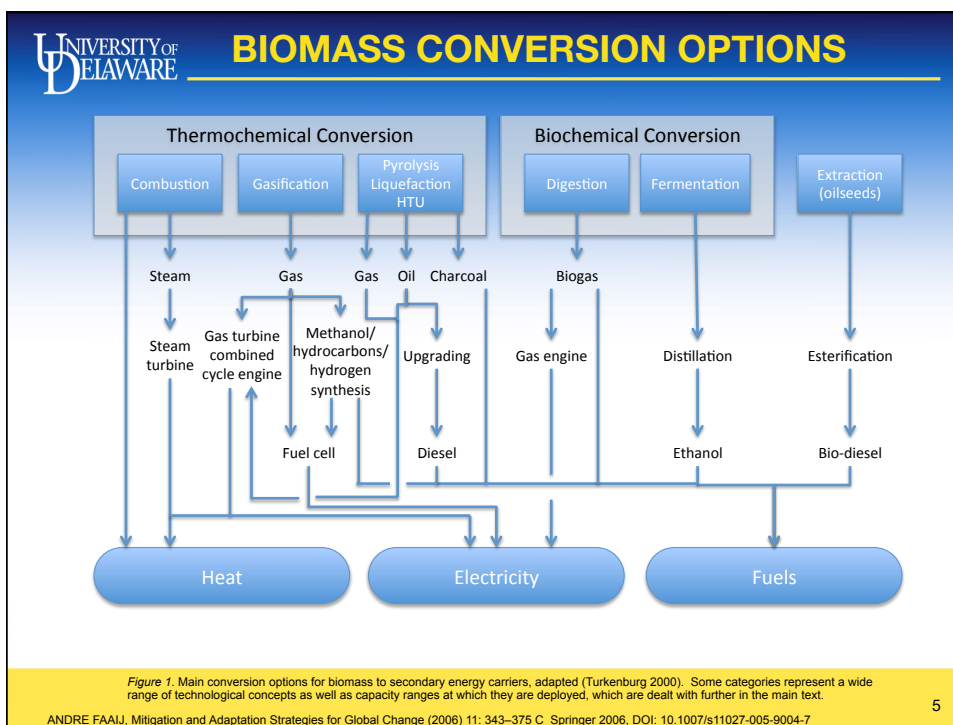
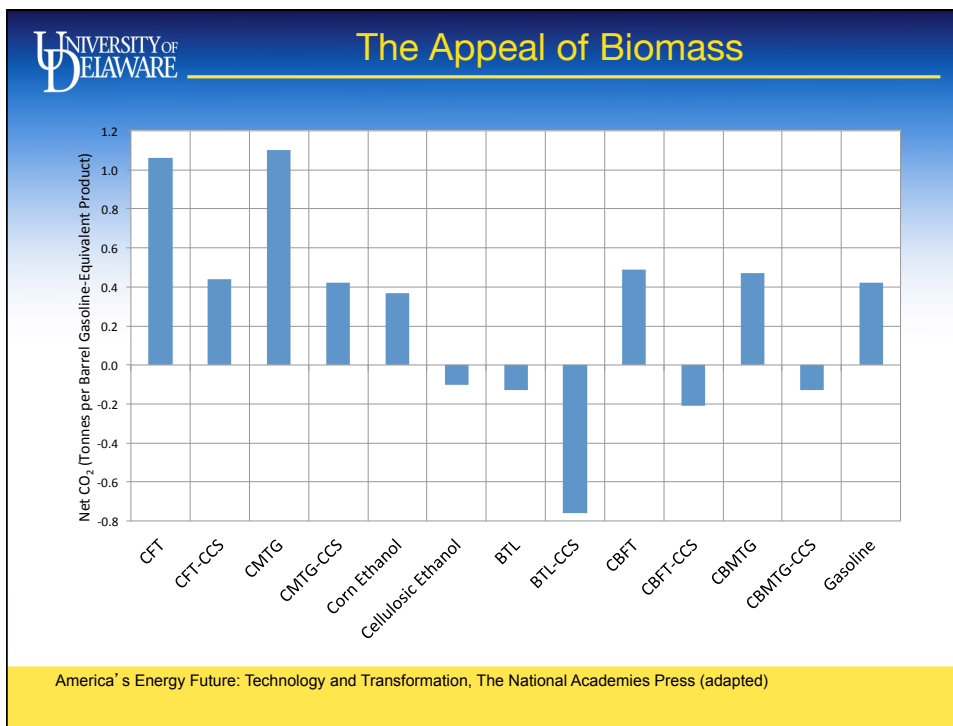
**Inaugural Fraunhofer - Delaware Technology Summit
Energy and Life Sciences —Solutions for Sustainability**
March 5, 2013
4:45-5:15 pm



Infrastructure-Ready Biomass-Derived Oils

Outline

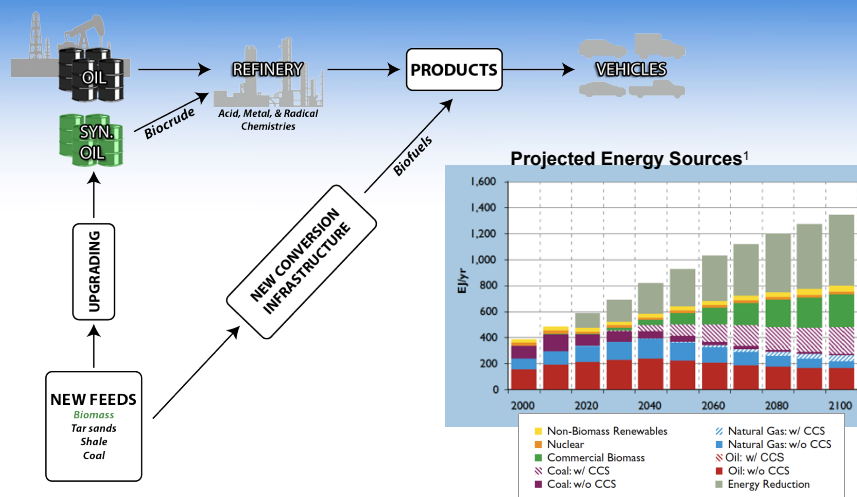
- The appeal of Biomass
- The significance of the infrastructure
- Infrastructure-ready options
- Reaction Engineering modeling tools to
“Organize, Evaluate and Improve”
Infrastructure-ready options
- Some preliminary modeling results



UNIVERSITY OF DELAWARE The Significance of Existing Infrastructure

- Valued at ~ 3-6 Trillion USD
- Investment decisions “deliberate” in nature
- Geographically-Based (will not relocate)
- Units with characteristic time of decades
- High bar for alternative “shut down” technology

UNIVERSITY OF DELAWARE Integration with Existing Infrastructure



[1] Clarke L.J., Edmonds H, et. al. *Scenarios of Greenhouse Gas Emissions and Atmospheric Concentrations*. 2007. U.S. Department of Energy.



Organizing, Evaluating and Improving

Integrated **modeling** of the reactions, process, economics and life-cycle analysis provides:

- Quantitative **organization** of **current** science and engineering knowledge base
 - The “discipline” of modeling
 - Quantitative models identify science gaps and needs
 - New models can be assembled in ~ 5 years
- Quantitative **evaluation** of **current** science and engineering knowledge base
 - Identifies needed improvements in technology and science
 - Precise research targets for issues that need to be addressed
- Quantitative **organization** of **improved** science and engineering knowledge base
 - Provides scenarios for the adoption of technology platforms
 - Objective technical basis for decision making

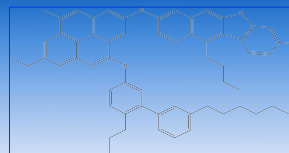
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Molecular-level Modeling Approaches and Tools

1. Structural and Composition Models

- Measurements (GS-MS, NMR, etc.)
- Modeling: transforming measurements to molecules
 - MolGen and Monte Carlo simulation of structure
 - CompGen: A “Fixed Footprint” approach



```
CCCCc1c2ccc(Sc3cc4cc(C)c5cc
(CC)cc6cc(Sc7ccc(CCC)c7)c7)
cccc(CCCCCC)c7)c(c3)c4c56)cc
2cc2sc3cc(C)ccc3c12
```

2. Reaction Modeling

- $N < N_C$: “Conventional” deterministic methods, e.g., INGen
- $N > N_C$: Stochastic methods, e.g., Monte Carlo simulation of reaction
- $N > N_C$: Hybrid methods, e.g., The Attribute Reaction Model

3. Reactivity Correlations

- Fundamentals based kinetics correlations for homologous family of reactions
- Order 10 [O(10)] LFERs for every process chemistry

4. Property Estimation


- Molecules to commercial lumps
- End-use vs. internal-use properties

	C_1	C_2	H_1	H_2	C_1	C_2	H_1	H_2			
C_1	0	1	0	0	Gasoline	Naphtha	Kerosene	LGO	DAGO	VGO	0
C_2	1	0	0	0	83.2	134.5	168.5	211.5	345.0	489.0	1
H_1	0	0	1	0	1.97	1.46	1.82	1.84	1.77	1.83	0
H_2	0	0	0	1	297	370	476	545	604	686	0
EBP(K)	302	437	479	545	546	653	723	793			
30BP(K)	310	460	517	546	653	723	793				
30BP(K)	318	460	517	546	653	723	793				
30BP(K)	326	461	517	546	653	723	793				
30BP(K)	334	461	532	584	690	805					
EBP(K)	358	461	544	588	693	810					

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Equivalent Computational Representations


 pentane

Bond Electron Matrix

	C ₁	C ₂	C ₃	C ₄	C ₅	H ₁	H ₂	H ₃	H ₄	H ₅	H ₆	H ₇	H ₈	H ₉	H ₁₀	H ₁₁	H ₁₂
C ₁	0	1	0	0	0	1	1	1	0	0	0	0	0	0	0	0	0
C ₂	1	0	1	0	0	0	0	0	1	1	0	0	0	0	0	0	0
C ₃	0	1	0	1	0	0	0	0	0	0	1	1	0	0	0	0	0
C ₄	0	0	1	0	1	0	0	0	0	0	0	0	1	1	0	0	0
C ₅	0	0	0	1	0	0	0	0	0	0	0	0	0	0	1	1	1
H ₁	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
H ₂	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
H ₃	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
H ₄	0	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
H ₅	0	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
H ₆	0	0	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0
H ₇	0	0	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0
H ₈	0	0	0	1	0	0	0	0	0	0	0	0	0	0	0	0	0
H ₉	0	0	0	1	0	0	0	0	0	0	0	0	0	0	0	0	0
H ₁₀	0	0	0	0	1	0	0	0	0	0	0	0	0	0	0	0	0
H ₁₁	0	0	0	0	1	0	0	0	0	0	0	0	0	0	0	0	0
H ₁₂	0	0	0	0	1	0	0	0	0	0	0	0	0	0	0	0	0

Adjacency List

Pentane

0 0 C: {1,1}, {5,1}, {6,1}, {7,1};
 1 0 C: {0,1}, {2,1}, {8,1}, {9,1};
 2 0 C: {1,1}, {3,1}, {10,1}, {11,1};
 3 0 C: {2,1}, {4,1}, {12,1}, {13,1};
 4 0 C: {3,1}, {14,1}, {15,1}, {16,1};
 5 0 H: {0,1};
 6 0 H: {0,1};
 7 0 H: {0,1};
 8 0 H: {1,1};
 9 0 H: {1,1};
 10 0 H: {2,1};
 11 0 H: {2,1};
 12 0 H: {3,1};
 13 0 H: {3,1};
 14 0 H: {4,1};
 15 0 H: {4,1};
 16 0 H: {4,1};

String Code

C(C(C(H3)H2)C(C(H3)H2)H2)

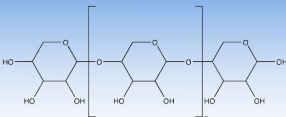
SMILES Code

CCCCC

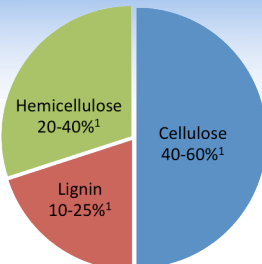
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Lignocellulosic Biomass



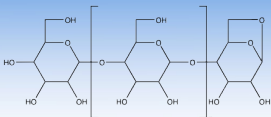
Hemicellulose
C₅H₈O₄



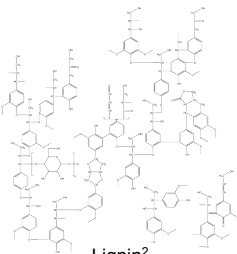
Cellulose
40-60%¹

Hemicellulose
20-40%¹

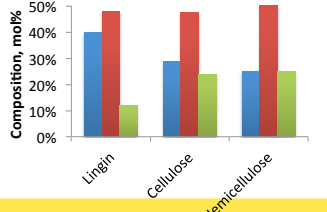
Lignin
10-25%¹



Cellulose
C₆H₁₀O₅



Lignin²
C₁₀H₁₂O₃



Composition, mol%

Component	C (%)	H (%)	O (%)
Lignin	~40	~48	~12
Cellulose	~30	~48	~22
Hemicellulose	~25	~48	~27

[1] Yang H, Yan R, et al. *Energy & Fuels*. 2006. 20 (1), 388-393.
 [2] Freudenberg K, Neish A.C. *Science*. 1969. 165 (3895), 784.

UNIVERSITY OF DELAWARE **Freudentberg Lignin Model as a String**

The computer “knows” Lignin as its SMILES code

CML Input from ChemDraw

```
O=Cc1cc(OC)c(c1)c1ccc(c1O)OC(C)OC2C1C
OC2c1ccc(c1)O)OC(C)c1ccc(c1)Occc1cc1O
C(C)(OC)ccc1C(C)(O)ccc(c1OC)C(C)(OC)ccc1
)C(C)(CO)C(CO)Occc(c1OC)C1c1ccc(c1)OC(=O)
CO(C)(CO)C(CO)CO)O)C(CO)OC(C1c1ccc(c1)OC)
O)Oc1(O)C)ccc1c1ccc(c1O)OC(C1C2C1=O)OC2C1C
c2c1ccc(c1)OC)O)C(C)(OC)ccc(c1OC)C1c1ccc(c1
cc1C1O=2(C1CO)ccc(c2OC)CC(=O)OC)OC(C)OC1(OC
)ccc1OC)C(C(CO)O)CO)CO)CO
```

Substituents: 39 Attributes
Cores : 19 Attributes
63 Connections:

12

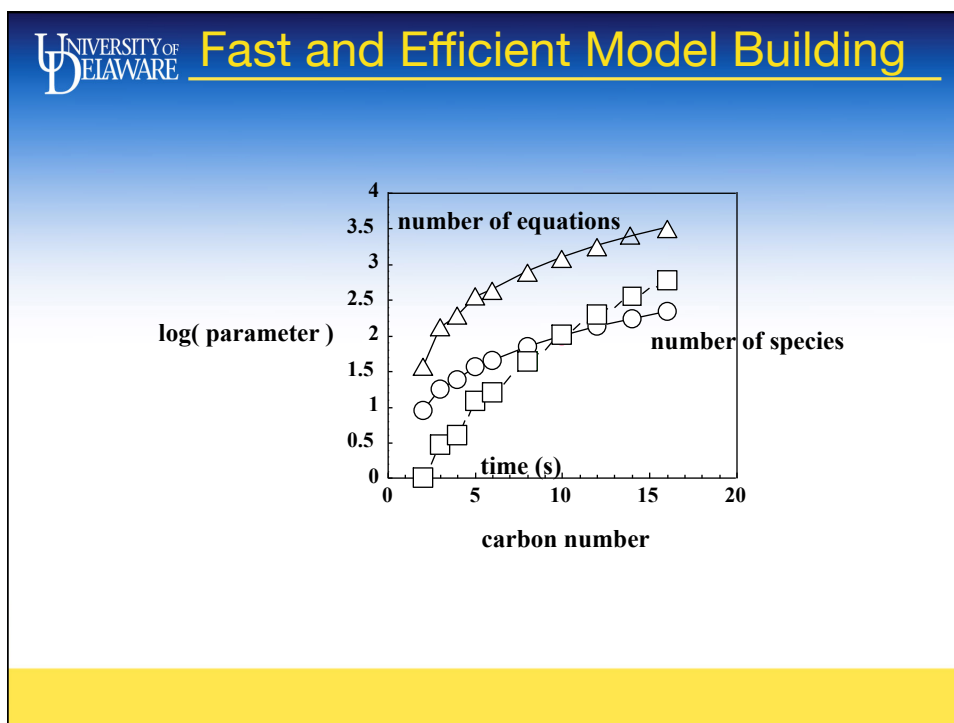
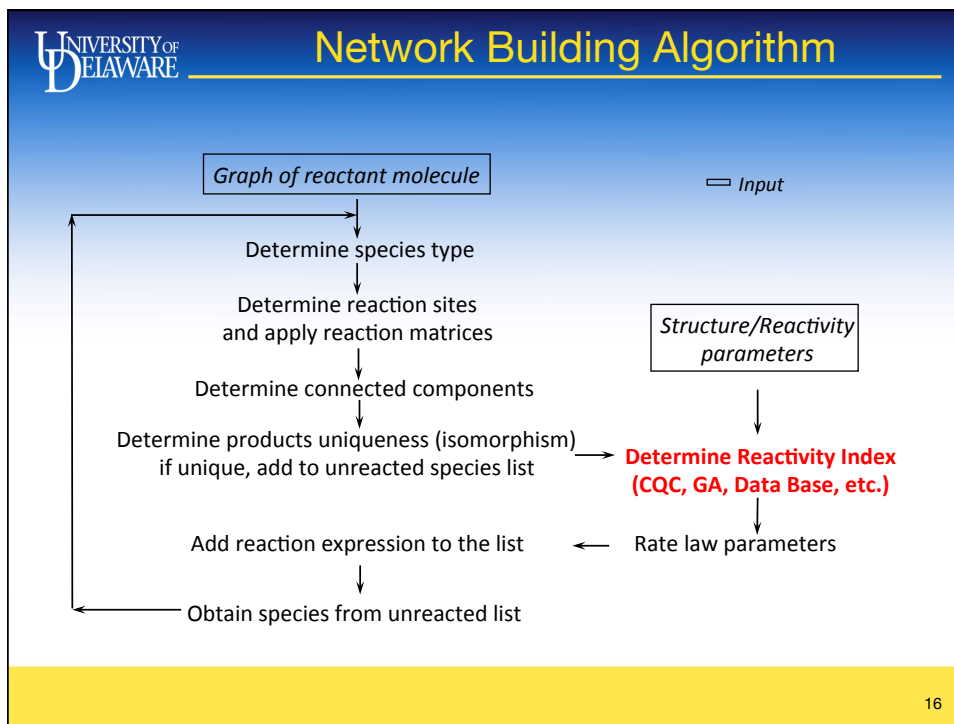
UNIVERSITY OF DELAWARE **Chemical Reaction as a Matrix Operation**

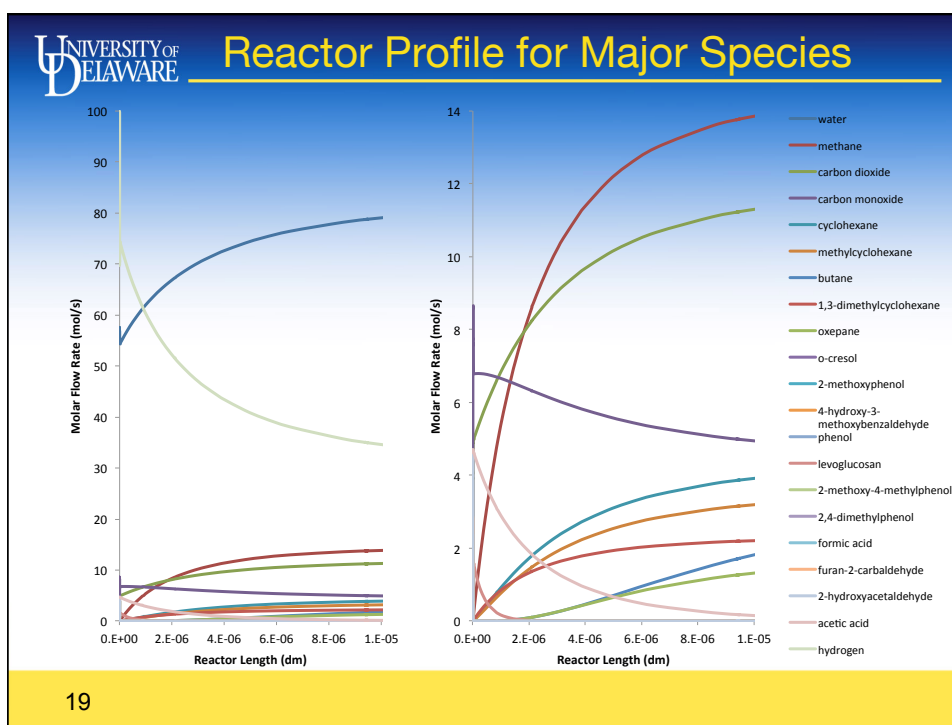
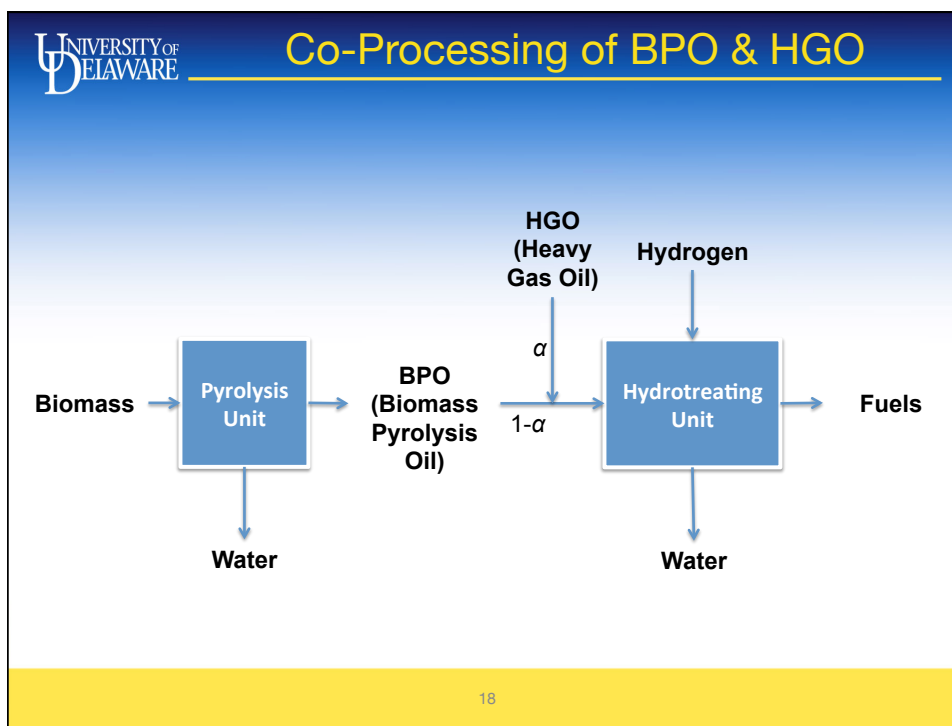
$$\begin{array}{c} \text{H} & \text{H} & \text{H} \\ | & | & | \\ \text{H}-\text{C}_1 & -\text{C}_2 & -\text{C}-\text{H} \\ | & | & | \\ \text{H} & \text{H} & \text{H} \end{array} + \text{H}_1-\text{H}_2 \rightleftharpoons \begin{array}{c} \text{H} \\ | \\ \text{H}-\text{C}_1-\text{H}_1 \\ | \\ \text{H} \end{array} + \begin{array}{c} \text{H} & \text{H} \\ | & | \\ \text{H}_2-\text{C}_2 & -\text{C}-\text{H} \\ | & | \\ \text{H} & \text{H} \end{array}$$

	C ₁	C ₂	H ₁	H ₂		C ₁	C ₂	H ₁	H ₂		C ₁	C ₂	H ₁	H ₂		
C ₁	0	1	0	0	+	C ₁	0	-1	1	0	=	C ₁	0	0	1	0
C ₂	1	0	0	0		C ₂	-1	0	0	1		C ₂	0	0	0	1
H ₁	0	0	0	1		H ₁	1	0	0	-1		H ₁	1	0	0	0
H ₂	0	0	1	0		H ₂	0	1	-1	0		H ₂	0	1	0	0

Reduced Reactant Matrix M_A Reaction Matrix M_R Reduced Product Matrix M_B

[1] Bennett CA. PhD Dissertation. Chemical and Biochemical Engineering, Rutgers University. 2009. (Adapted)





Organize, Evaluate and Improve

Preliminary Status

- Quantitative organization of science and engineering knowledge base
- Identified some key technology gaps, science needs
 - Petroleum was inherently profitable, which allowed for 100+ years of technology development
 - Biomass not inherently competitive, issues need to be addressed
 - H₂ consumption: expensive
 - H₂ consumption: CO₂ footprint
 - H₂O release
 - O rejection as H₂O or CO₂?
 - Energy density

Modeling Conclusions

- Irreducible complexity, as measured by model parameters:
 - Structure: O(10) 5-10 PDFs × 2-3 parameters each
 - Reactivity: O(10) 10 LFERs × 2 parameters each
 - Reaction Families: O(10) Reaction Matrices
- Rigorous kinetics can model interactions and competitions
- Net reduction of complexity: O(10⁵ × 10⁵) parameters reduced to O(30)